

Capacitive field-effect (bio-)chemical sensors based on nanocrystalline diamond films

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ABSTRACT

Capacitive field-effect electrolyte-diamond-insulator-semiconductor (EDIS) structures with O-terminated nanocrystalline diamond (NCD) as sensitive gate material have been realized and investigated for the detection of pH, penicillin concentration, and layer-by-layer adsorption of polyelectrolytes. The surface oxidizing procedure of NCD thin films as well as the seeding and NCD growth process on a Si-SiO₂ substrate have been improved to provide high pH-sensitive, non-porous thin films without damage of the underlying SiO₂ layer and with a high coverage of O-terminated sites. The NCD surface topography, roughness, and coverage of the surface groups have been characterized by SEM, AFM and XPS methods. The EDIS sensors with O-terminated NCD film treated in oxidizing boiling mixture for 45 min show a pH sensitivity of about 50 mV/pH. The pH-sensitive properties of the NCD have been used to develop an EDIS-based penicillin biosensor with high sensitivity (65-70 mV/decade in the concentration range of 0.25-2.5 mM penicillin G) and low detection limit (5 µM). The results of label-free electrical detection of layer-by-layer adsorption of charged polyelectrolytes are presented, too.

INTRODUCTION

Artificially grown diamond is a promising transducer material for chemical and biological sensing, as it is widely considered as biocompatible, displays outstanding electrical and electrochemical properties, and allows the direct coupling of biomolecules onto the diamond surface [1-5]. Among the various proposed concepts for the development of diamond-based chemical sensors and biosensors, the semiconductor field-effect platform is one of the most attractive approaches. Most of diamond-based field-effect (bio-)chemical sensors reported have been realized on a transistor structure by using hydrogen (H)-terminated polycrystalline or monocrystalline diamond films as an active transducer material [6-8].

Owing to the simplicity of the layout, the absence of a complicated encapsulation procedure and thus, an easier and cost-effective fabrication, capacitive field-effect structures are especially suited for (bio-)chemical sensor applications. Therefore, recently, we have introduced a capacitive field-effect electrolyte-diamond-insulator-semiconductor (EDIS) structure as a platform for (bio-)chemical sensing [9-11]. This work summarizes recent experimental results on the development of EDIS sensors for the detection of pH, penicillin concentration and layer-by-layer adsorbed charged macromolecules using an oxygen (O)-terminated nanocrystalline diamond (NCD) films as transducer material.

EXPERIMENTAL

Undoped NCD thin films of ~100 nm thickness were grown on a p-Si-SiO₂ ($\rho=1-10 \text{ }\Omega\text{cm}$, 50 nm thermally grown SiO₂) structure by means of microwave (2.45 GHz) plasma-enhanced chemical vapor deposition from a mixture of methane (CH₄) and hydrogen (H₂) in an ASTeX 6500 reactor. Prior to growth, the SiO₂ surface was seeded with a monodisperse colloid of nanocrystalline diamond particles in water with an ultrasonic bath. The NCD growth process on a SiO₂ as well as an additional surface treatment in oxidizing medium have been optimized to yield high pH-sensitive non-porous O-terminated NCD films without damage of the underlying SiO₂ layer. An Al film was deposited on the rear side of the Si chip as a contact layer. The chip size of the EDIS sensors has been 10 x 10 mm². For more details about the preparation of NCD films, see [12,13].

Typically, as prepared NCD surfaces are hydrogen (H)-terminated. In order to obtain O-terminated surfaces, the NCD films were treated in an oxidizing mixture of H₂SO₄ and KNO₃. The NCD films have been physically characterized by means of scanning electron microscopy (SEM), atomic force microscopy (AFM) and X-ray photoelectron spectroscopy (XPS) methods. The SEM micrograph in Figure 1 exemplarily demonstrates the surface morphology of a 100 nm thick NCD film. The film comprised randomly orientated fine grains and was totally closed. No visible pores in the film or damages of the underlying SiO₂ layer have been observed.

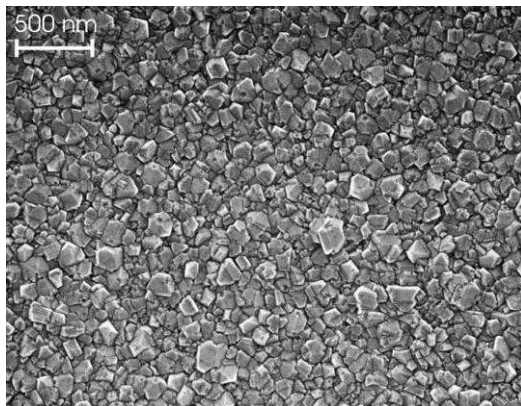


Figure 1. SEM picture of the surface morphology of a 100 nm thick NCD film grown on a p-Si-SiO₂ structure.

Additional AFM analysis of the diamond films was performed in intermittent contact mode (BioMat Workstation, JPK Instruments - Germany) and revealed a nanostructured surface consisting of NCD grains having a size of ~100 nm with an average surface roughness of 11 nm as depicted in Figure 2.

The EDIS sensors have been electrochemically characterized in buffer solutions with different pH values by means of capacitance-voltage and constant-capacitance (ConCap) method using an impedance analyzer (Zahner Elektrik). For the experiments, the EDIS sensors were mounted into a home-made measuring cell, sealed by an O-ring and contacted on its front side by the electrolyte and on its rear side by a gold-plated pin. A conventional liquid-junction Ag/AgCl electrode was used as a reference electrode. For more detailed information on the measurement set-up, see [9].

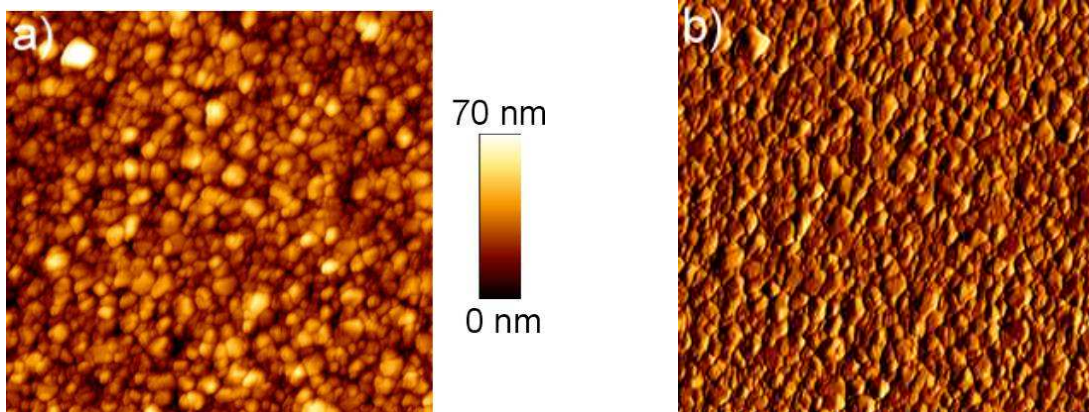


Figure 2. AFM height (a) and phase (b) images of a NCD film. The scan size is $2 \times 2 \mu\text{m}^2$.

RESULTS AND DISCUSSION

EDIS-based pH sensor

The pH sensitivity of O-terminated EDIS sensors has been studied by means of ConCap method. Figure 3 demonstrates exemplarily a typical dynamic pH-response (a) and calibration curve (b) of an EDIS structure with a 100 nm thick O-terminated NCD film oxidized by wet chemical oxidation for about 45 min. The ConCap signal has been recorded in Titrisol buffer solutions with different pH values from pH 4 to pH 12. The O-terminated EDIS sensors displayed a pH sensitivity of 48-50 mV/pH. The pH-sensitive properties of NCD films can be explained by the site-binding model, similar to ion-sensitive field-effect transistors. The surface potential of the diamond film could be influenced by the presence of hydroxyl groups at the oxidized diamond surface, resulting in the pH-dependent modulation of the space-charge capacitance in the semiconductor and the sensor output signal.

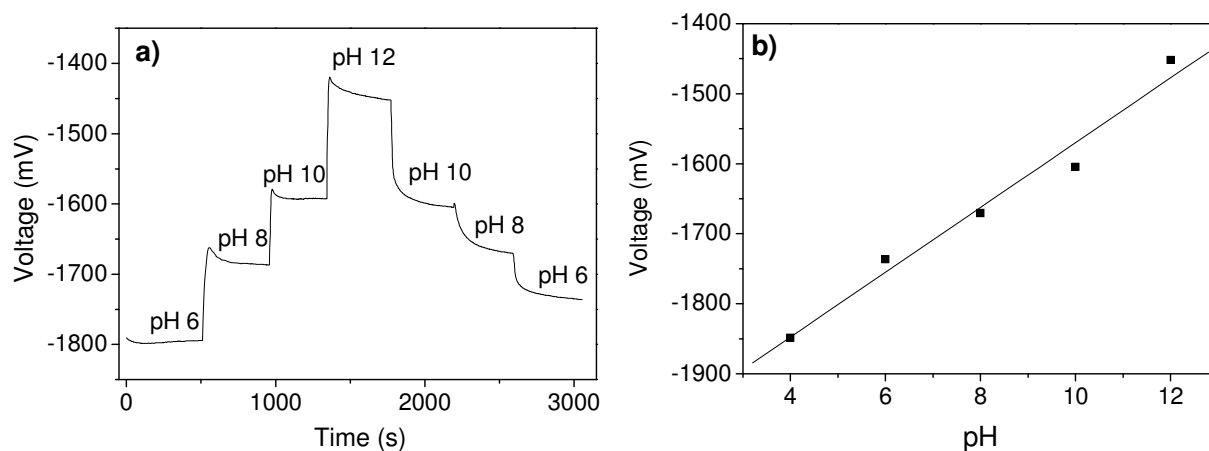


Figure 3. Typical ConCap response (a) and calibration curve (b) for O-terminated NCD film recorded in Titrisol buffer of different pH values from pH 6 to pH 12.

The obtained values of pH sensitivity of NCD films are slightly lower than for Al₂O₃, Si₃N₄, and Ta₂O₅ that have often been used as pH-sensitive transducer materials for field-effect devices [14]. The main advantages of NCD films are their chemical inertness and biocompatibility as well as the possibility of interfacing to biological systems by direct coupling of biomolecules onto the diamond surface via carbon groups.

EDIS-based penicillin biosensor

The high pH-sensitive properties of the EDIS structures have been used to develop a penicillin biosensor. The EDIS penicillin biosensor detects the local pH change near the surface as a result of the catalyzed hydrolysis of penicillin by the enzyme penicillinase. Figure 4 demonstrates an example of the dynamic ConCap response of the developed EDIS-based penicillin biosensor. In this experiment, the enzyme penicillinase (EC 3.5.2.6., *Bacillus cereus* from Sigma, specific activity: 1650 units/mg protein) was adsorptively immobilized directly onto the O-terminated NCD surface. The measurements have been performed in 0.5 mM polymix buffer solutions (pH 8, 100 mM KCl as an ionic-strength adjuster) with different penicillin concentrations from 5 μ M to 2.5 mM. The penicillin solutions were prepared by dissolving penicillin G (benzyl penicillin, 1695 units/mg, Sigma) in the working buffer.

With increasing penicillin concentration, the concentration of the H⁺ ions resulting from the penicillin hydrolysis is increased, too. As a result, the voltage that is necessary in order to adjust the constant capacitance raises. The freshly prepared NCD-based penicillin biosensor possessed a low detection limit of 5 μ M and a high sensitivity of 65-70 mV/decade in the concentration range from 0.25 to 2.5 mM penicillin G.

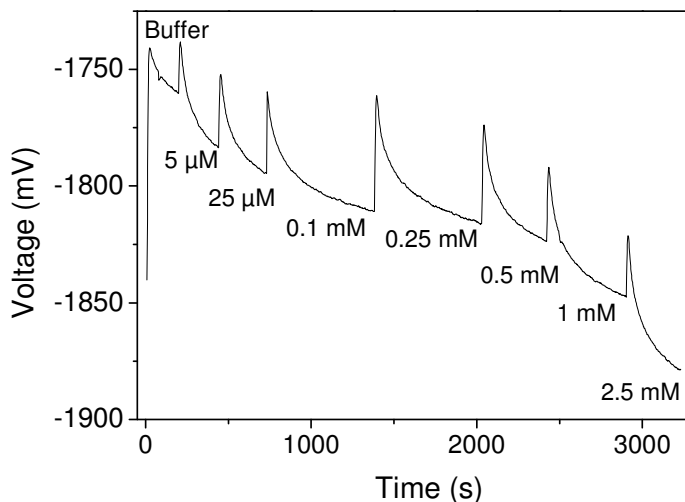


Figure 4. Typical ConCap response of the developed NCD-based penicillin biosensor.

Label-free detection of charged macromolecules with EDIS sensor

In order to demonstrate functional capabilities of NCD films for multi-sensing applications, the realized EDIS structures with O-terminated NCD film have been used for a label-free electrical detection of charged macromolecules. The positively charged PAH (Poly

(allylamine hydrochloride)) and negatively charged PSS (Poly (sodium 4-styrene sulfonate)) polyelectrolytes (PE) were chosen as model system. The multilayers of PAH/PSS were obtained by using the layer-by-layer assembly technique (see e.g., [15-17]) by sequential adsorption of PAH and PSS from the respective PE solution (50 μ M PAH or PSS, pH 5.4). During the experiment, the EDIS sensors were consecutively exposed to the respective PE solution for a time necessary for the adsorption of each single monolayer, followed by rinsing in electrolyte solution. We started the formation of the PE multilayer onto O-terminated NCD with positively charged PAH.

Alternating potential changes, having the tendency to decrease with increasing the number of adsorbed polyelectrolyte layers (from 35-40 mV for first PE layers to 2-4 mV for an EDIS structure with 14-15 PE layers), have been observed after the adsorption of each PE layer. The adsorption of negatively charged PSS shifts the sensor signal towards the direction of a more negative surface charge, whereas the adsorption of the positively charged PAH shifts the sensor signal into the direction corresponding to a more positive charged NCD surface. Thus, the molecular layers induce an interfacial potential change resulting in alternating changes in the flat-band voltage, and therefore, in the output signal of the EDIS structure.

CONCLUSIONS

The presented results demonstrate the potential of field-effect capacitive EDIS structures having NCD films as transducer material for multi-parameter sensing of different (bio-)chemical quantities. The immobilization of other biomolecules, like DNA, proteins, on NCD films for extending the biosensor capabilities will be subject of future research.

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