

Surface plasmon resonance investigation of gold nanoparticle aggregation on self-assembled monolayers

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Motivation

pH sensor applications

→ Bulky size

- pH electrodes ^[1]



Ambient medium $\Lambda \wedge \bullet$ Au film Prism





Signal enhancement of pH-sensitive polymers ^[5] by AuNP



Results

I) SPR sensor surface with P2VP-SH and AuNP-COOHs



 $-P2VP-AuNP(\sigma = 3.2 \%)$ -P2VP

pH response of P2VP-SH with ($\sigma = 3.2$ %) and without AuNP-COOHs. Enhanced decrease SPR signal due to swelling on the surface with AuNP-COOHs.



3.2 % coverage (σ) of Ø20 nm **AuNP-COOHs on P2VP-SH**



Kinetics of RI change on P2VP-SH SAM with AuNP-COOHs.

Highly sensitive towards high pH values, pH 11. AuNP-COOHs repulsion on the shrunk polymer result in a strong decrease in SPR signal.

II) SPR sensor surface with 4-mercaptopyridine (MP) (have no swelling effect) and AuNP-COOHs.



Aggregation and disaggregation of AuNP-COOHs with different coverage.



Conclusion & Outlook

Conclusion

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- Swelling of polymers with AuNP-COOH at pH 2.3 to 1.7 (decrease in SPR signal)
- Distance increase between AuNPs and Au substrate (weaker Localised SPR coupling)

References

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- Shrinking of polymers at pH 11 and AuNP-COOH disaggregation
- Strong decrease in SPR signal due to AuNP-COOH repulsion



Electrostatic repulsion of AuNP-COOHs accompanied by shrinkage of the P2VP chains at pH > 7. II. At pH = 7, the P2VP chains are partly swollen and the AuNPs can approach each other. **III.** Aggregation of the AuNP-COOHs accompanied by the complete swelling of P2VP chains at pH < 7.

Outlook

- Different AuNP size, shape (octahedral, rod, triangle) and with other functionalities (-NH₂)
- Different pH-sensitive polymers with acidic functional group (-COOH)
- Microfluidic sensor system

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Acknowledgement

First of all, I appreciate the chance that Prof. Wagner provided for me and connect me to such professional group. I would like to thank Prof. Mertig for giving me the opportunity to work in TU Dresden and KSI. I also would like to thank my supervisors, Dr. Kick and Dr. Lakatos for their useful ideas and endless support throughout my master thesis.