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Radiological evaluation of industrial residues for construction purposes correlated with their chemical properties

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Abstract

This study characterises the naturally occurring radionuclide (NOR) contents of a suite of secondary raw materials or industrial residues that are normally disposed of in landfills or lagoons but now are increasingly used in green concretes. This includes ashes from a variety of industrial processes and red mud from aluminium production, as well as air pollution control residue and cement kiln dust. The chemical composition of the samples was determined with X-ray fluorescence spectroscopy (XRF). The Ra-226, Th-232 and K-40 activity concentrations were obtained by gamma spectrometry, and the results were compared with recently published NOR databases. The correlation between the NOR contents and the main chemical

22 composition was investigated. The radioactive equilibrium in the U-238 chain was studied based on the
23 determination of progeny isotopes. The most commonly used calculation methods (activity concentration
24 index and radium equivalent concentration) were applied to classify the samples. The radon exhalation rate
25 of the samples was measured, and the radon emanation coefficient was calculated. Significant correlation
26 was found between the NORs and certain chemical components. The massic exhalation demonstrated a
27 broad range, and it was found that the emanation coefficients were significantly lower in the case of the
28 residues generated as a result of high-temperature combustion processes. The results showed a weak
29 correlation between the Ra-226 concentration and the radon exhalation. This emphasizes that managing the
30 Ra-226 content of recycled material by itself is not sufficient to control the radon exhalation of recycled
31 materials used in building products. The investigated parameters and their correlation behaviour could be
32 used to source apportion materials found during the process of landfill mining and recovery of material for
33 recycling.

34

35 Keywords: gamma spectrometry; radon emanation; radon exhalation; recycling; red mud

36 Highlights

- 37 • The NOR contents of the surveyed residues fit well with recent NOR databases
- 38 • The radioactive equilibrium in the decay chains fits well with the literature data
- 39 • Strong correlation was found between the NOR contents and certain chemical elements
- 40 • The chemical composition can be used as a source apportionment tool for NOR content
- 41 • The emanation was low for residues generated under high-temperature processes

42

43 1. Introduction

44 The depletion of raw materials and demand for low carbon material resources has resulted in the urgent
45 need for new eco-innovative building materials. The market need for efficient, economical and safe
46 production of new building products requires a comprehensive model of the properties of primary and
47 secondary raw materials, and existing building products to facilitate the creation of a circular economy.
48 *“The transition to a more circular economy, where the value of products, materials and resources is*
49 *maintained in the economy for as long as possible, and the generation of waste minimised, is an essential*
50 *contribution to the EU's efforts to develop a sustainable, low carbon, resource efficient and competitive*
51 *economy”* (European Commission, 2015). These material resources are often recycled from secondary raw
52 materials or industrial residues rather than from pristine sources. Valorisation alongside carbon capture
53 technologies can offer an opportunity for a circular economy approach to building materials (Pan et al.,
54 2018). Alkali-activated materials (AAMs) are alternative low-carbon binders and can be produced through
55 the reuse of industrial residues as secondary raw materials. These residues have traditionally been disposed
56 of in brownfield landfills and lagoons which are now being mined to recover the valuable materials.
57 These residues often serve as precursors for production of AAM. Ideally the activators should be also
58 residues or secondary raw materials with high pH (Tong et al., 2018). Here we focus on the residues that
59 can be used in the production of AAMs. Globally, about 46% of CO₂ emissions originated as a result of the
60 fossil fuel combustion, with 31% emitted from coal combustion power plants (Olivier et al., 2016). The
61 amount of generated residue depends on the non-combustible mineral content, which can vary between 5-
62 30% (Kovacs et al., 2017). As a result of combustion, the NOR contents of the coal are enriched in the
63 residues (Table 1). A tenfold enrichment factor, compared to the initial NOR contents of the coal, is common
64 in residues (Somlai et al., 1996). Owing to this phenomenon, residues, including the coarse bottom ash
65 (BA), also called coal slag, and the fine, fly ash (FA), contain elevated (Kovacs et al., 2017; Kardos et al.,
66 2015) and occasionally extremely high concentrations of Ra-226 (Somlai et al., 2006; Somlai et al., 1996)
67 above the 1000 Bq/kg clearance level defined in the EU BSS (European Union, 2014). The use of BA and

68 FA in construction materials always increases the risk to residents. The iron production process produces
69 slag as a residue. The NOR contents of the ground granulated blast-furnace slag (GGBFS) (Table 1) depend
70 on the materials used during the process such as iron ore, sinter, flux limestone, dolomite, and coke.
71 Aluminium production from bauxite produces red mud. The red mud still contains valuable compounds.
72 Huge efforts are made to find economical technology to valorise them. However, when reused on an
73 industrial scale it will still be subject to an ‘end of waste criteria’ depending on the country where this
74 occurs. Globally, this is estimated at 150 million tons per year (Davris et al., 2015). The safe reuse of red
75 mud has not been resolved yet, due to its heavy metal and its elevated NOR contents (Table 1). The unsafe
76 storage of the red mud can endanger eco-systems and humans, as was demonstrated by disasters that
77 happened in 2010 at Ajka, Hungary (Somlai et al., 2010; Gelencsér et al., 2011; Mayes et al., 2016) and in
78 2016 at Luoyang, Henan province, China, the latter fortunately without human injury (Liu, 2016). Red mud
79 needs to be checked for its NOR contents when it is used in building materials (Kovacs et al., 2017). Cement
80 kiln dust (CKD) presents a significant landfill and disposal problem (Gunning et al., 2010). There was no
81 information found in the current literature regarding the NOR contents of the CKD. However, for
82 comparison, cement data can be used as a proxy with the assumption that the CKD generated from the same
83 stream as cement production is composed of the same unreacted raw material (US EPA, 2016). Municipal
84 residue, clinical residue and sewage sludge produce significant amounts of ash (Rani et al., 2008). The ash,
85 e.g. household waste ash (HWA), incinerated sewage sludge ash (ISSA) and dust, e.g. air pollution control
86 (APC) residue, originate as a result of the scrubbing of emissions, with the residue often classified as
87 hazardous (Kourti et al., 2010; Rani et al., 2008). Numerous studies have dealt with their reuse in the cement
88 and concrete industry (Cyr et al., 2007; Donatello and Cheeseman, 2013; Wongsa et al., 2017; Müller and
89 Rübner, 2006). As a result of the incineration, the radionuclides mainly remain in the solid residues
90 (Carvalho, 2017). A study performed in Germany (Puch et al., 2005) measured the NOR contents of the
91 HWA and compared them with the world averages published in RP-112 (European Commission, 1999).
92 The radionuclide contents of ISSA mainly depend on the treated wastewater. The NORs and also artificial
93 radionuclides, most notably I-131, Tl-201, and Sr-89 (all short half-lived medical isotopes), can accumulate

94 in the sludge and after the incineration remain in the inert inorganic particles of the ISSA. A survey,
95 involving 313 Publicly Owned Treatment Works across the USA, concluded that elevated levels of
96 radioactive materials were found in some sewage sludge and ash samples, but the survey did not indicate a
97 widespread problem (ISCORS, 2005). Rice husk ash (RHA), a residue of biomass-based electricity
98 generation, originates from rice mills. The husk consists of approximately 40% cellulose, 30% lignin and
99 20% silica and is burned in power plants producing a residue with a hard, abrasive nature. The high silica
100 content (~90-95%) and the reactive nature of the amorphous silica content makes it usable as a pozzolana
101 (He et al., 2013). Information in the scientific literature about the NOR contents in RHA is currently missing.
102 This circular approach requires cross-disciplinary collaboration between academics, industry, and the
103 authorities. Large quantities of specific secondary raw materials can be utilised in AAMs as solid binders
104 of performance comparable to Portland cement (Ascensão et al., 2017; Bondar and Coakley, 2014; Puertas
105 et al., 2015; Vinai et al., 2016). For new forms of concrete, the use of AAMs incorporating industrial
106 residues reduces CO₂ emissions by up to 80% (Aiken et al., 2017). Although economically viable, market
107 uptake and widespread application of these new types of recycled materials are currently hampered by public
108 health concerns related to immobilisation of potentially toxic compounds (Sas and Vandevenne, 2015) and
109 in particular radiation exposure from naturally occurring radionuclides (NORs) that can be present in these
110 materials (Kovacs et al., 2017).

111 With existing construction materials and raw materials, the contents of NOR (U-238, Th-232 series and
112 their progenies, K-40) and their effects can have significant health implications. Despite the relatively low
113 NOR contents generally found in buildings, the radionuclides present can cause long-term exposure to the
114 inhabitants, due to prolonged indoor residence times. Two main exposure pathways can be differentiated
115 when considering the built environment. The primary pathway leading to internal exposure is as a result of
116 the decay of the incorporated radionuclides from the U-238 and Th-232 decay series that leads to the
117 formation of radon & thoron. The exposure factor, related to this pathway, is the inhalation of radon/thoron
118 (Rn-222, Rn-220) and their progenies (from soil and construction materials) which can get stuck in the lungs
119 and irradiate cells. Radon, a radioactive noble gas, can diffuse out of the building materials and is currently

120 the second most common cause of lung cancer after smoking (Axelsson et al., 2015; World Health
121 Organization, 2009). Depending on the internal structure and the parent element concentration (Ra-226),
122 the amount of radon which can exhale from a given building material matrix can vary greatly. The secondary
123 pathway, the external exposure caused mainly by gamma radiation emitted from the decay of NOR also
124 needs to be considered. Owing to their high penetration capacity, high energy gamma photons can exit walls,
125 floors, and ceilings that contain NOR materials. This depends on several factors including the energy of the
126 gamma photons, the thickness of the walls, the material density, the homogeneity and concentration of the
127 radionuclides. A constant dose rate is formed as a result of the NORs in buildings. The gamma exposure
128 can also depend on the position of the residents within rooms, but typically to a small extent (International
129 Atomic Energy Agency, 2012; Risica et al., 2001). Screening of potential industrial residues prior to
130 recycling ensures the avoidance of recycled building materials with increased radioactivity, and, in this way,
131 the industrial residue streams for the circular economy are checked in advance to assure safe recycling
132 (Kovacs et al., 2017).

133 To avoid an elevated risk for residents in houses, the EU has laid down Council Directive 2013/59/Euratom
134 (European Union, 2014) as Basic Safety Standards (EU-BSS) for protection against the dangers arising from
135 exposure to ionising radiation. Annex VIII of the directive defines the requirement for maximum allowable
136 excess dose from the NOR contents of building materials. Furthermore, the reference level for indoor radon
137 was set at 300 Bq/m³ average radon concentration. However, uniform requirements and a standardised
138 method for screening the radon release from building materials before they are placed on the market are still
139 missing, which presents a significant challenge for future research. The NOR contents of construction
140 materials, their raw materials and residues depend on the origin of the materials. The processing technology
141 can also lead to a strong fluctuation in the NOR concentration, even in the case of the same residue repository
142 (Croymans et al., 2017b).

143 Table 1 presents global information about the NOR contents of BA, FA, GGBFS, HWA and RM based on
144 the NORM4Building (Schroeyers et al., 2018) and By-BM (By-Products for Building Materials) (Sas et al.,

145 2017) NOR databases. In relation to the other residues that this article considers, namely APC, CKD, ISSA,
 146 RHA, there are no records yet in the NOR databases (Sas, 2017; Sas et al., 2017; Schroeyers et al., 2018).

147

148 **Table 1: Naturally occurring radionuclide (NOR) contents (Bq/kg) of studied residues**
 149 **according to NORM4Building and By-BM databases**

Material	Database	Ra-226			Th-232			K-40		
		Min	Max	Average	Min	Max	Average	Min	Max	Average
BA	By-BM	16	3152	845	11	290	845	7	1100	253
FA	NORM4B	11	1000	188	1	200	91	17	1100	343
	By-BM	14	1028	235	1	250	96	44	3001	505
GGBFS	NORM4B	100	323	201	25	148	66	158	500	298
	By-BM	8	399	183	3	330	83	7	388	162
HWA	By-BM	11	25	18	8	21	13	159	213	83
RM	NORM4B	97	1700	389	45	1800	553	15	583	216
	By-BM	97	1047	311	118	1350	324	5	583	155

150

151 There is a fundamental difference between the two NOR databases. The By-BM database operates with
 152 individually reported sample information, including information on how the samples were measured, which
 153 enables further statistical analysis of the data. The disadvantage of this dataset is that it operates with a
 154 limited amount of records (currently a few thousand). The dataset contains the number of the samples with
 155 the minimum, maximum, and the average or median activity concentrations of samples. The
 156 NORM4Building dataset has a greater number of records but consists mainly of averaged data. This type of
 157 information does not allow the precise weighting of the data, but the great number of entries provides an
 158 overall insight into the materials.

159 Here we evaluate new radiological and chemical analysis of residues or secondary raw materials commonly
160 used in the manufacture of alkali-activated materials by determination of NOR contents by gamma
161 spectrometry and calculation of I-index and radium equivalent. We investigate radioactive disequilibrium
162 in the radioactive decay chain of Th-232 and U-238 and determine massic exhalation and radon emanation
163 factor for the residues.

164 2. Materials and methods

165 2.1 Sample preparation

166 Samples were collected from different European countries and originate from various industrial activities
167 (Table 2). The sources of the samples are not revealed, as the present scientific research is not for judging
168 them. The samples were homogeneous within each source, represented by a unique Sample ID in Table 2.
169 FA 7 was originated from the same site as FA 1 and GGBFS 2 came from the same site as GGBFS 1. This
170 is the reason why FA 7 and GGBFS 2 were not examined with XRF. Fig. 1 gives exact compositions of all
171 these sample materials. All ashes are siliceous. Note that the data shown in Table 1 are from literature data
172 forming the two databases, while the samples listed in Table 2 are for the original experimental studies, the
173 results of which are reported in this paper. The samples were dried to constant mass in a drying cabinet at a
174 temperature of 105°C and powdered. A typical sample weight used for subsequent analysis post drying was
175 1 kg. The powdered samples were divided to three portions. For the gamma spectrometry analysis and the
176 radon exhalation measurement, typically 0.3–0.5 kg samples were used respectively depending on their
177 density. For the XRF characterization 10 g samples were used.

178

Table 2: Identifier (ID) of surveyed materials

Sample ID	Material	Industry
APC	Air pollution control residue	Municipal waste incineration
BA	Bottom ash	Coal combustion thermal power plant
CKD 1, 2, 3	Cement kiln dust	Cement production
FA 1, 2, 3, 4, 5, 6, 7	Fly ash	Coal combustion thermal power plant
GGBFS 1, 2	Ground granulated blast-furnace slag	Ferrous industry
HWA	Household waste ash	Municipal waste incineration
ISSA	Incinerated sewage sludge ash	Sewage sludge incineration
RHA	Rice husk ash	Rice milling industry
RM 1, 2	Red mud	Aluminum production

180

181 **2.2 Determination of major components with XRF**

182 In this study, the main chemical composition was determined by X-ray fluorescence (XRF) analysis. The
 183 measurements were performed on a fused glass bead sample, for removing both grain size and mineralogical
 184 effects. The samples were heated under air atmosphere up to 1200°C in a platinum crucible with lithium
 185 borates. During the process, oxides of the metal contents are formed and the results are reported in that form.
 186 The equipment used was a PANalytical Axios Advanced XRF spectrometer which runs on a 4KW Rh tube
 187 using WDS and a sample to flux ratio of 1:10. The results, which were quoted as weight percent, were
 188 analysed using PANalytical SuperQ software using reference samples and artificial analogues. LOI was
 189 determined by igniting the materials at 950°C for 1.5 hours.

190 **2.3 Determination of NOR with gamma spectrometry**

191 The samples were put into polystyrene containers with metal cap and stored for 27 days to achieve secular
 192 equilibrium between Ra-226 and Rn-222. The NOR contents of the investigated materials were determined

193 by gamma spectrometry using a calibrated high purity broad energy germanium detector (Canberra BE5025-
 194 7500SL) with a 50% nominal relative efficiency. The detector was shielded with a copper-lined lead shield
 195 specific for low-activity measurements. All samples were measured for 80 000 s using the same poly(methyl
 196 methacrylate) sample holder to fix the samples in position 5 mm above the endcap of the detector, leaving
 197 an air-filled gap between the sample and the detector. Canberra's Genie 2000 software was used for data
 198 acquisition. Canberra's LabSOCS software was used to perform the efficiency calibration with self-
 199 absorption correction. The 1 sigma error was calculated by the software. The determined radionuclides and
 200 their details are presented in Table 3.

201

202 **Table 3: Details of radionuclides measured by gamma spectrometry**

Nuclide	Decay chain	Determined isotope	Energy (keV)	Intensity (%)
Pb-214	U-238	Ra-226	351.9	35.6
Bi-214	U-238	Ra-226	609.3	45.5
Th-234	U-238	U-238	63.3	3.8
Pb-210	U-238	Pb-210	46.5	4.3
Ac-228	Th-232	Th-232	911.2	26.2
Pb-212	Th-232	Th-232	238.6	43.6
K-40	-	K-40	1460.8	10.6
Cs-137	-	Cs-137	661.7	85.0

203

204 **2.4 Calculation of commonly used indexes**

205 2.4.1 Radium equivalent index

206 The radium equivalent index (Ra_{eq}) (Beretka and Mathew, 1985) is based on the assumption that the dose
 207 rate contribution of unit activity of Ra-226, Th-232 and K-40 with their belongings decay chains are

208 different. The dose criterion in the case of Ra_{eq} is 1.5 mGy/year absorbed dose which corresponds to a value
 209 of 1.0 mSv annual effective dose. The calculation of Ra_{eq} assumes that 259 Bq/kg of Th-232 and 4810 Bq/kg
 210 of K-40 cause a dose rate equivalent to 370 Bq/kg of Ra-226 (Beretka and Mathew, 1985; Nuccetelli et al.,
 211 2017). The Ra_{eq} can be calculated according to the following equation:

$$212 \quad Ra_{eq} = A_{Ra-226} + 1.43 A_{Th-232} + 0.077 A_{K-40} \quad (1)$$

213 where A_{Ra-226} , A_{Th-232} , and A_{K-40} are the activity concentration of Ra-226, Th-232, and K-40, respectively.
 214 The current Chinese and Russian legislation is derived from this approach (Nuccetelli et al., 2017).

215 2.4.2 Activity concentration index (I-index)

216 To characterise construction materials the calculation of the I-index is one of the most commonly used
 217 screening tool based on the Mikka Markannens model, presented in RP112 (European Commission, 1999).
 218 The I-index is derived to indicate whether the annual dose due to the excess external gamma radiation in a
 219 building may exceed 1.0 mSv. This calculation method operates with the assumption that all the walls, the
 220 ceiling and also the floor of the room are made from 20 cm thick concrete with 2350 kg/m³ density. The
 221 computation considers the occupancy factor, the annual indoor spent time, and the dose conversion factor.
 222 The targeted 1.0 mSv dose excess can be the result of exposure to respectively 276 Bq/kg Ra-226, 231
 223 Bq/kg Th-232 or 3176 Bq/kg of K-40. In the final formula of the I-index, the values computed above are
 224 rounded to the nearest full 100 Bq/kg (Ra-226 and Th-232) or 1000 Bq/kg (K-40). The latest European
 225 Basic Safety Standard (EU-BSS) in Article 75, Annex XIII (European Union, 2014) introduces a screening
 226 index which is based on the abovementioned computation to identify building materials that are of concern
 227 from the radiological protection point of view. The I-index can be calculated according to the following
 228 formula (European Commission, 1999; Nuccetelli et al., 2017):

$$229 \quad I = \frac{C_{Ra}}{300Bq/kg} + \frac{C_{Th}}{200Bq/kg} + \frac{C_K}{3000Bq/kg} \quad (2)$$

230 where C_{Ra} , C_{Th} , C_K are the Ra-226, Th-232 and K-40 activity concentrations expressed in Bq/kg.

231 In the European Union, the member states were required to harmonise their national legislation according
232 to the EU-BSS at the latest in February 2018. It is important to note that the calculation of the I-index only
233 allows for a conservative screening. To more accurately predict the dose contribution as a result of recycled
234 building materials in a given building, the thickness and the density also have to be taken into consideration
235 (Nuccetelli et al., 2015). A recently published dose model (Croymans et al., 2017a) with expanded gamma
236 lines from NORs is applicable in non-standard rooms.

237 **2.5 Massic radon exhalation and emanation factor of surveyed samples**

238 The massic exhalation (Friedmann et al., 2017) of surveyed samples was determined with an accumulation
239 chamber technique. The granulated (<10 mm grain size) (Friedmann et al., 2017), dry samples were
240 enclosed in radon-tight acrylic accumulation chambers equipped with stainless steel valves. To effectively
241 remove the radon from the samples, including pores, the chambers were evacuated in a vacuum chamber
242 and after purged with intensive radon-free airflow. The exhaled radon was determined with RD200 (200
243 cm³ chamber volume) ionisation chamber manufactured by FTLab. Circulation was used for 10 minutes
244 with 5 L/min flow rate. To avoid the disequilibrium of the radon and its progenies, and the contribution of
245 thoron (Jonas et al., 2016), the first result obtained from the 60 minutes of the measurement was ignored. In
246 the accumulation time function, the activity of the decay of the exhaled radon was corrected, and the massic
247 exhalation was calculated according to the following formula (Sas et al., 2015b):

$$248 \quad E_{Mass} = \frac{C_t \cdot V}{m \cdot t} \cdot \frac{\lambda \cdot t}{1 - e^{-\lambda t}} \quad (3)$$

249 where C_t = accumulated radon concentration in the measurement kit during sampling [Bq m^{-3}], E_{Mass} =
250 massic exhalation rate [$\text{mBq kg}^{-1} \text{h}^{-1}$], t = accumulation time [h], V = volume of the accumulation kit [m^3],
251 m = mass of the sample [kg], λ = decay constant of radon [h^{-1}]. Error values were derived from the sensitivity
252 of the instrument, using information provided by manufacturer.

253 The emanation factors of the surveyed samples were calculated from the Ra-226 activity concentration
254 obtained by gamma spectroscopy and the equilibrium radon concentration calculated from the massic
255 exhalation results (International Atomic Energy Agency, 2013):

$$256 \quad \varepsilon = \frac{Rn_{eq}}{C_{Ra-226}} \cdot 100 \quad (4)$$

257 where ε = emanation factor (%), Rn_{eq} = equilibrium radon concentration and C_{Ra-226} = Ra-226 activity
258 concentration.

259 3. Results and discussion

260 3.1 Chemical composition of surveyed samples obtained by XRF

261 The main chemical compositions of the studied residues are illustrated as a heat-map in Fig. 1. Dark red
262 represents the dominant components in the samples. The SiO₂ content was the highest with 90.8 wt% in the
263 case of the rice husk ash sample (RHA) which fits the reported data (He et al., 2013). In the case of those
264 fly ash and bottom ash samples, the SiO₂ ranged between 42.7-56.0 wt%, the Al₂O₃ content varied between
265 18.4-30.0 wt% (highest among all residues studied) and the Fe₂O₃ ranged between 4.9-18.7 wt%. The fly
266 ash samples were all siliceous and not calcareous. The CaO content was dominant in the case of all CKD,
267 APC and GGBFS 1 samples (32.6-49.5 wt%). In the case of both red mud samples, the composition fits
268 well with the scientifically reported data (Wang and Liu, 2012). The most dominant components in the red
269 mud were Fe₂O₃ (35.8-43.8 wt%, highest among all residues studied), Al₂O₃ (16.3-25.1 wt%), SiO₂ (8.6-
270 13.8 wt%), TiO₂ (5.1-10.2 wt%, highest among all residues studied), CaO (4.5-5.6 wt%) and Na₂O (4.7-7.7
271 wt%) which mainly originates from the NaOH solution used in the Bayer process (Kovacs et al., 2017).

Sample ID	SiO ₂	TiO ₂	Al ₂ O ₃	Fe ₂ O ₃	MnO	MgO	CaO	K ₂ O	Na ₂ O	P ₂ O ₅	SO ₃	V ₂ O ₅	Cr ₂ O ₃	SrO	ZrO ₂	BaO	NiO	CuO	ZnO	PbO	LOI
APC	6.43	0.79	2.59	0.73	0.08	1.05	32.59	3.55	3.20	0.85	16.38	0.01	0.03	0.05	0.01	0.06	0.00	0.07	1.78	0.05	14.40
BA	56.02	0.83	18.39	7.97	0.09	1.73	5.36	2.61	0.88	0.68	0.23	0.03	0.01	0.18	0.04	0.34	0.01	0.01	0.01	0.00	4.63
CKD 1	16.37	0.21	3.73	2.21	0.06	0.88	46.62	4.35	0.37	0.13	13.67	0.01	0.01	0.07	0.01	0.04	0.00	0.01	0.06	0.03	9.90
CKD 2	14.57	0.27	3.84	2.07	0.06	0.75	49.53	9.12	1.07	0.09	4.28	0.01	0.00	0.02	0.00	0.02	0.01	0.05	0.04	0.05	13.17
CKD 3	11.83	0.24	3.49	2.39	0.18	1.60	45.44	8.29	0.93	0.05	4.61	0.06	0.01	0.03	0.01	0.02	0.01	0.03	0.01	0.02	20.21
FA 1	50.24	0.98	25.84	11.27	0.08	1.86	3.15	2.91	0.89	0.22	0.12	0.08	0.02	0.10	0.03	0.32	0.02	0.02	0.03	0.01	3.05
FA 2	53.64	0.91	20.31	9.26	0.07	1.83	4.33	2.19	1.14	0.43	0.39	0.04	0.01	0.18	0.04	0.33	0.01	0.01	0.02	0.00	4.54
FA 3	48.70	0.98	29.97	6.99	0.07	1.45	2.80	3.26	0.66	0.68	0.58	0.05	0.02	0.14	0.04	0.42	0.03	0.03	0.02	0.01	3.97
FA 4	55.45	1.07	23.90	4.86	0.05	1.72	4.43	1.89	0.81	0.72	0.73	0.03	0.01	0.18	0.00	0.43	0.01	0.01	0.02	0.01	3.19
FA 5	55.77	0.83	19.84	5.79	0.08	2.03	4.46	2.42	1.24	0.53	0.55	0.03	0.01	0.15	0.04	0.35	0.01	0.01	0.02	0.01	5.44
FA 6	42.73	0.91	27.90	18.72	0.04	0.98	4.13	1.45	0.11	0.40	0.71	0.04	0.02	0.10	0.03	0.25	0.01	0.01	0.03	0.01	2.18
GGBFS 1	34.74	0.62	13.80	0.34	0.46	7.63	41.13	0.69	0.30	0.00	2.54	0.01	0.00	0.08	0.04	0.42	0.00	0.00	0.00	0.00	0.00
HWA	35.22	0.89	12.47	10.57	0.16	2.39	19.59	1.07	4.21	1.74	1.43	0.06	0.05	0.16	0.04	0.33	0.04	0.24	0.48	0.07	7.34
ISSA	34.19	1.26	13.53	15.02	0.19	3.46	9.21	2.36	1.25	16.98	0.31	0.02	0.03	0.05	0.03	0.23	0.02	0.10	0.33	0.02	1.46
RHA	90.78	0.07	1.16	0.71	0.14	0.54	0.78	2.00	0.08	1.28	0.04	0.00	0.00	0.00	0.01	0.07	0.00	0.00	0.01	0.00	2.59
RM 1	8.58	10.17	16.25	43.76	0.05	0.07	5.64	0.09	4.68	0.32	0.16	0.23	0.25	0.01	0.35	0.21	0.01	0.00	0.01	0.01	9.33
RM 2	13.75	5.07	25.13	35.78	0.02	0.41	4.54	0.57	7.72	0.03	0.39	0.09	0.10	0.00	0.13	0.76	0.06	0.00	0.01	0.02	5.25



272

273 **Fig. 1:** Main chemical components (wt%) of studied residues illustrated on a heat-map chart

274 **3.2 NOR contents**

275 The activity concentration values obtained with gamma spectrometry of the NOR and the Cs-137 contents
 276 are presented in Table 4. Compared to the world average of NORs in construction materials (Ra-226 = 50
 277 Bq/kg, Th-232 = 50 Bq/ kg and K-40 = 500 Bq/kg) (European Commission, 1999), the obtained average
 278 values of the residues were 1.84, 1.8 and 1.5 times higher, respectively.

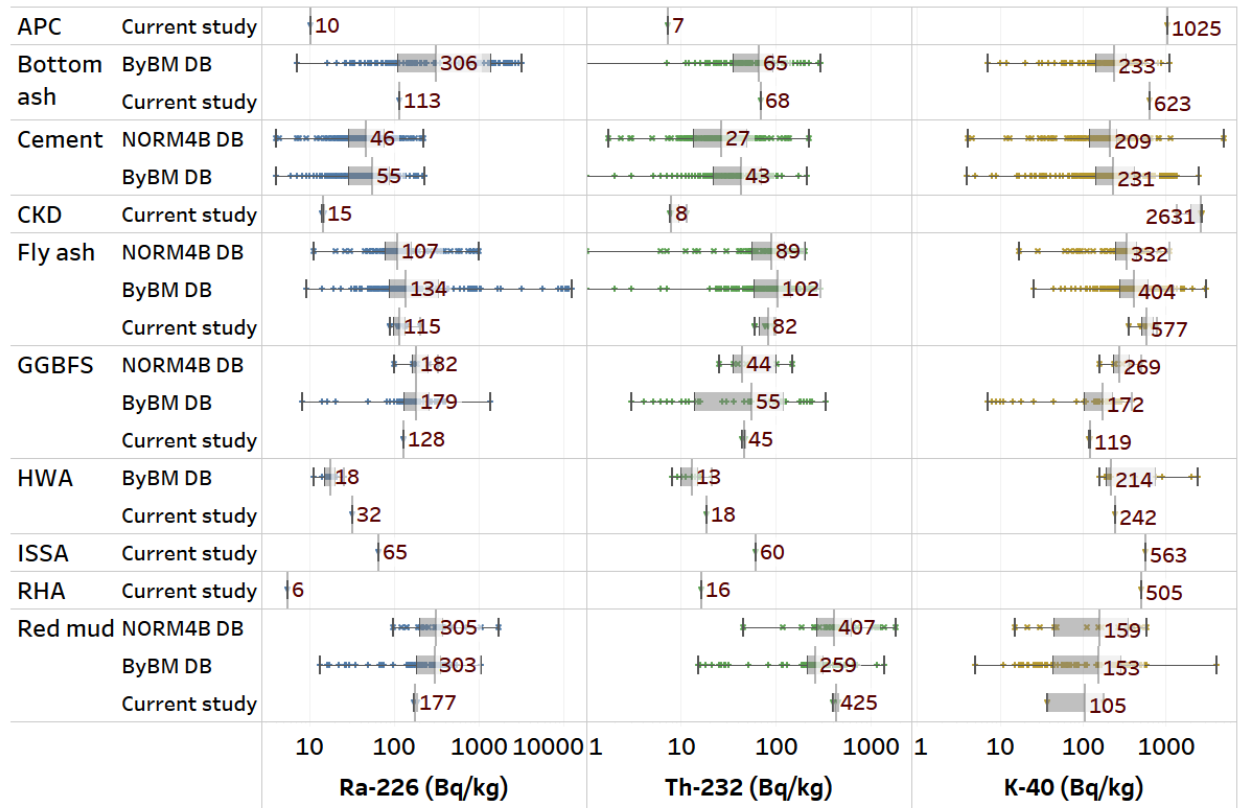
279 There is no easy way to compare the different materials. They are associated with different raw materials
 280 and different processes. The NOR isotopes are trace elements and their behavior depends on the processes.
 281 These materials are completely independent from each other. It is not a straightforward mechanism in
 282 general. So, the results were compared in this paper only to the records of the NORM databases. A source
 283 apportionment tool can be a first step to build a dataset to provide possibility to draw conclusion for the
 284 mechanisms.

285 **Table 4: Naturally occurring radionuclide (NOR) and Cs-137 contents of surveyed materials**
 286 **where the 1 sigma error was calculated by the LabSOCS software**

ID	Ra-226	Th-232	K-40	Cs-137
APC	10±1	7±1 (min.)	1025±43	6.1±0.4
BA	113±8	68±5	623±27	-
CKD 1	15±1	11±1	1348±57	0.9±0.1
CKD 2	15±1	8±1	2712±113 (max.)	4.9±0.4
CKD 3	15±1	8±1	2631±110	3.0±0.3
FA 1	139±10	82±6	743±31	1.4±0.2
FA 2	108±7	59±4	542±23	-
FA 3	89±6	94±7	763±32	-
FA 4	115±8	98±7	482±21	-
FA 5	89±6	60±4	577±25	-
FA 6	201±14 (max.)	97±7	349±15	-
FA 7	129±9	76±5	674±29	1.3±0.1
GGBFS 1	126±9	44±3	117±6	-
GGBFS 2	129±9	47±3	122±6	-
HWA	32±2	18±1	242±11	1.4±0.1
ISSA	65±5	60±4	563±24	11±1 (max.)
RHA	6±1 (min.)	16±1	505±22	-
RM 1	186±13	452±31 (max.)	37±2 (min.)	-
RM 2	168±11	398±27	174±8	-
Average	92	90	749	2

287

288 In Table 4, a dash line indicates no detectable data, i.e., the measured intensity was below the detection
 289 limit. The results of different studied materials were compared with the similar material records in the
 290 NORM4Building (Schroeyers et al., 2018) and By-BM (Sas et al., 2017) NOR databases. The result of the
 291 comparison is illustrated in Fig. 2.



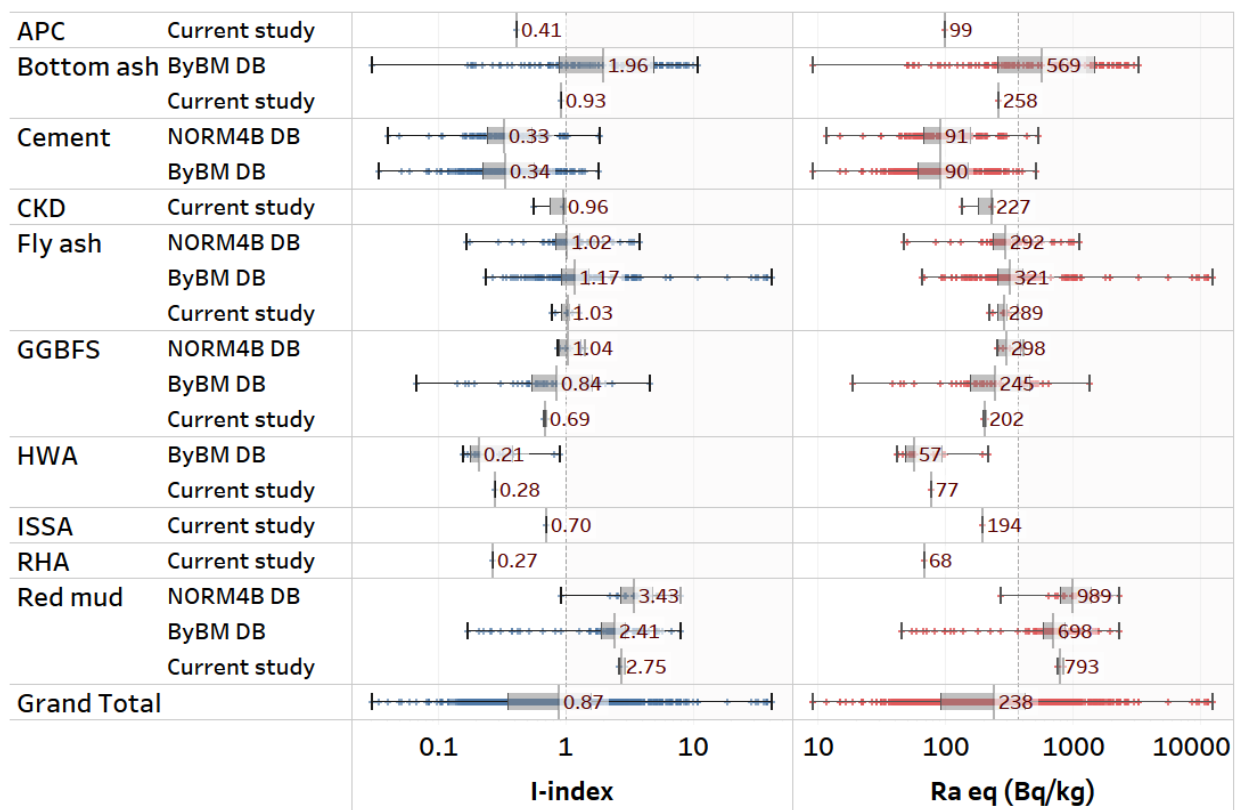
292
 293 **Fig. 2:** Comparison of obtained naturally occurring radionuclide (NOR) contents with datasets of NOR
 294 databases with the distribution of the data and the computed average values all illustrated

295 Regarding APC, CKD, ISSA and RHA samples, a comparison was not possible because there was no
 296 information about these materials in either database. In the case of the other surveyed residues, their NOR
 297 contents were lower than the averages in the databases except that all three investigated NOR of the HWA,
 298 the K-40 contents of the bottom ash and fly ash samples and the Th-232 contents of the red mud were found
 299 above the average values. However, since there was no information about the CKD in the databases, the
 300 obtained results were compared with the cement records. In the case of the CKD results, the Ra-226 and

301 Th-232 contents were significantly lower than the average values of the cement records collected in the
 302 databases. The K-40 contents were significantly higher compared to the database records.

303 3.3 I-index and radium equivalent concentration of studied residues

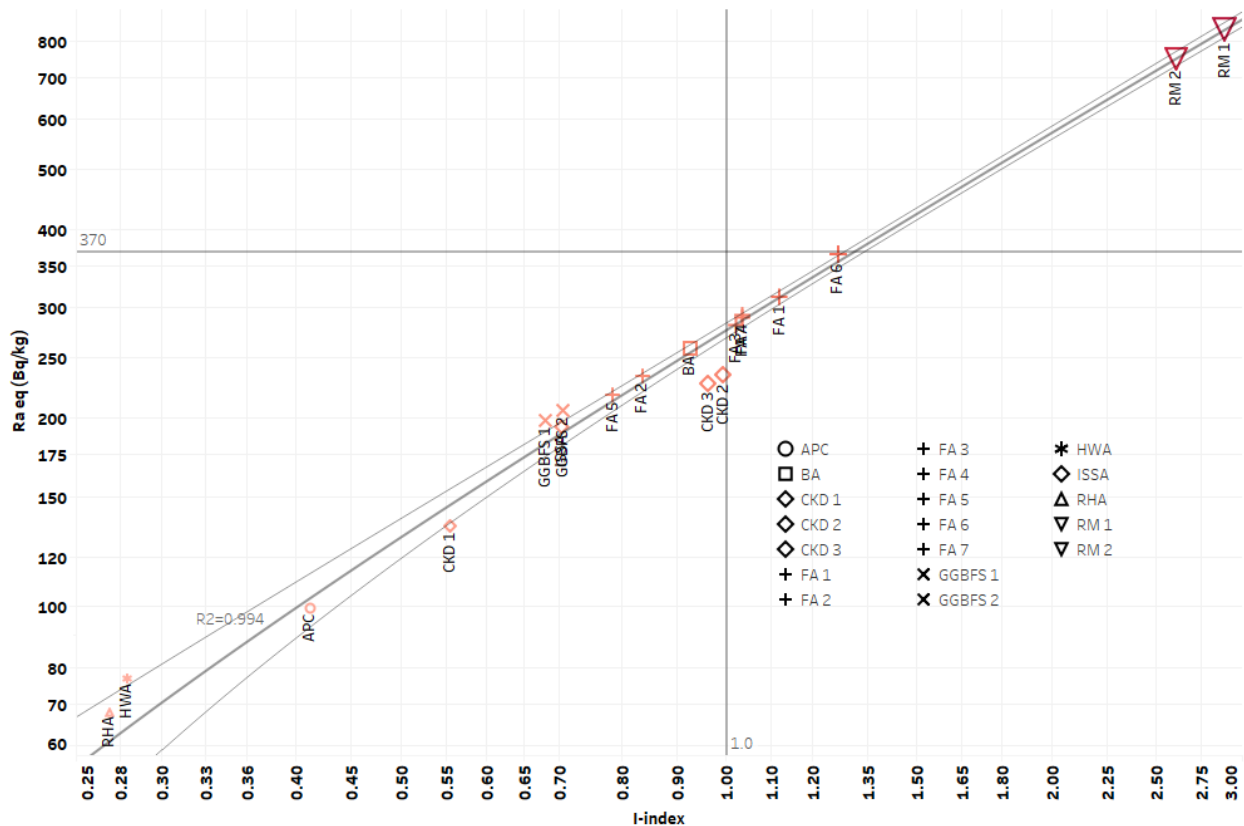
304 The I-index and the radium equivalent concentration were calculated based on the obtained NOR content.
 305 However, this calculation provides only a conservative screening value, since neither the density nor the
 306 thickness of the construction materials are taken into consideration. The results are illustrated in Fig. 3.



307
 308 **Fig. 3:** Activity concentration index (I-index) and radium equivalent index (Ra_{eq}) concentration of
 309 surveyed residues compared with naturally occurring radionuclide (NOR) datasets

310 The I-index values varied between 0.27-2.89 with an average of 1.00. The I-index of five siliceous fly-ash
 311 and two red mud samples were found with above the EU recommended 1.0 I-index value (European Union,
 312 2014). The Ra_{eq} concentration of the residues was lower than 370 Bq/kg for all cases, except the red mud
 313 samples. However, the Ra_{eq} concentration seems to be considerably more permissive compared to the I-

314 index: it has to be highlighted that 370 Bq/kg Ra_{eq} concentration is equivalent to 1.2 times the I-index value
 315 which is about 20% higher than the I-index recommended value of 1.0. In the case of the I-index, the targeted
 316 excess external dose from the building materials is 1.0 mSv/a. The dose contribution of the ICRP 115
 317 (Tirmarche et al., 2010) and EU-BSS (European Union, 2014) recommended 300 Bq/kg average radon
 318 concentration which corresponds to approximately 10 mSv/year effective dose (Paquet et al., 2017). Clearly,
 319 the acceptable excess dose from the gamma exposure is significantly lower relative to the acceptable radon
 320 exposure. The comparison of the calculated values is presented in Fig. 4. The trendline was placed with its
 321 95% confidence bands to represent the uncertainty in an estimate of a curve or function based on limited or
 322 noisy data. As confidence intervals are constructed and only refer to a single point, they are narrower (at
 323 this point) than a confidence band which is supposed to hold simultaneously at many points (Härdle et al.,
 324 2004).

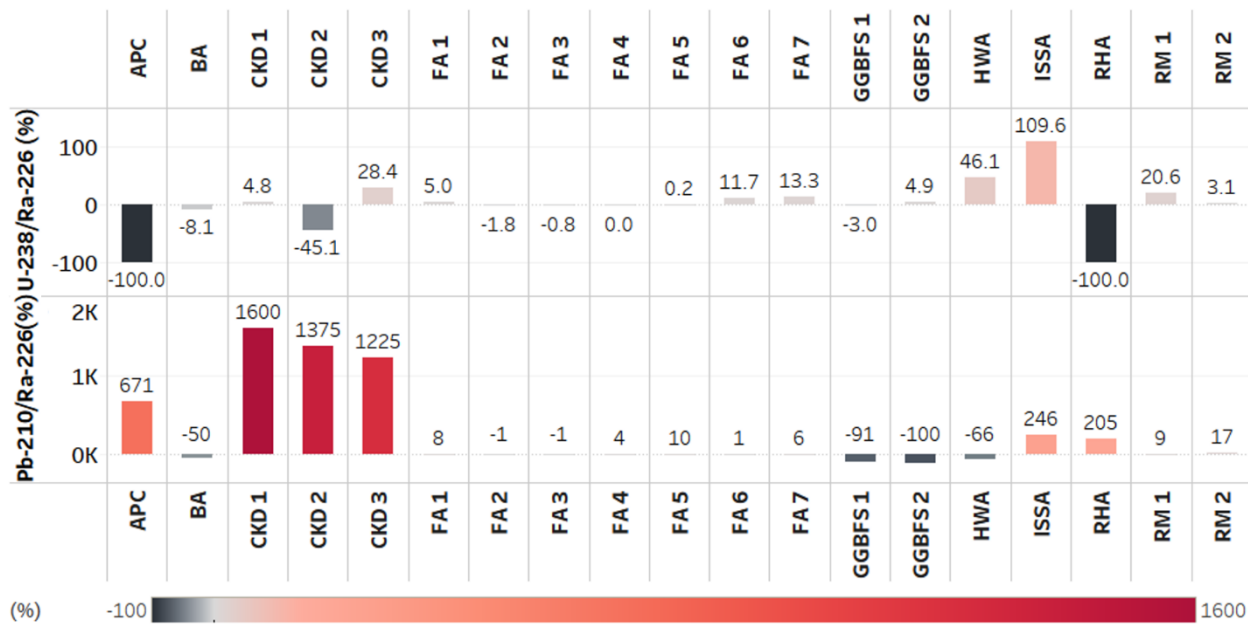


325
 326 **Fig. 4:** Relationship between the calculated I-indexes and radium equivalent concentrations of studied
 327 residues with 95% confidence bands

328
329 However, certain samples do not fulfil the criteria of I-index. It has to be noted that both the Ra_{eq} and the I-
330 index are used for only screening of construction materials. Individually, none of the studied residues are
331 suitable for direct use as a 100% building material. This means the final I-index and the Ra_{eq} of any building
332 material that contains a proportion of these residues have to be calculated to incorporate the residues and
333 other components of any building material based on their total NOR contents (Sas et al., 2017; Schroeyers
334 et al., 2018). According to the databases, the aggregates and the cement used for concrete production have
335 significantly lower NOR contents resulting in a lower I-index. The mixing of the industrial residues with
336 other construction material components (aggregates/cements) dilutes the NOR contents and allows the final
337 NOR contents of the recycled construction materials to be under levels defined by different national
338 legislation (Sas et al., 2015a; Nuccetelli et al., 2017).

339 **3.4 Radioactive equilibrium of U-238 and Th-232 decay chains of the studied** 340 **industrial residues**

341 The radioactive equilibrium state of U-238 decay chain was investigated. The U-238 content was obtained
342 from the Th-234 content (63.3 keV gamma line) of the studied samples. The Pb-210 activity concentrations
343 were obtained via the 46.5 keV gamma peak. To check the equilibrium condition in the chain, the Ra-226
344 activity concentrations were used as a reference value. However, in the case of the Th-232 chain, the most
345 significant possibility for disequilibrium may occur between the Th-232 and Ra-228 owing to the 5.74 years
346 half-life of Ra-228. Since direct measurement of Th-232 via gamma and alpha spectrometry was not
347 possible, its content was not determined. There was no information about the date when the residues were
348 generated so exact age of the samples was also not known. Owing to these facts, the disequilibrium state
349 was not investigated in the case of Th-232 chain. The disequilibrium state of U-238 chain related to Ra-226
350 content is presented in Fig. 5.



351

352

Fig. 5: Equilibrium state of U-238 decay chain related to Ra-226 content

353

The calculated values showed that the U-238 disappeared in the case APC and the RHA samples and in the

354

case of CKD 2 the U-238 content was 45.1% lower compared to Ra-226 activity concentration. The

355

disequilibrium state of RHA suggests that the uranium uptake or bioaccumulation by the rice from soil is

356

extremely low. Significant U-238 enhancement was found in the case of HWA and ISSA samples. Both

357

samples originate as a result of incineration of materials containing significant amounts of various organic

358

compounds. Pb and the Po are relatively volatile elements in high-temperature combustion environments as

359

is the case for coal combustion, cement production, and incineration of wastes. In this study, all the samples

360

are generated from high-temperature treatment, except for red mud, which is produced by the Bayer-process.

361

Owing to the high temperature, the volatile radionuclides condensate on the fine ash particles (Ozden et al.,

362

2018) which can result in elevated levels of Pb-210 in the fly ash type residues with decreased levels in the

363

bottom ash residues. The bottom ash residues, the BA, and both GGBFS and HWA samples had significantly

364

lower Pb-210 content compared to their Ra-226 content. The fine ash type residues had significantly

365

increased Pb-210 content compared to Ra-226 except for all FA samples. The Pb-210 content of all CKD

366

samples was extremely high with 13-16 times higher Pb-210 content compared to the Ra-226 activity

367

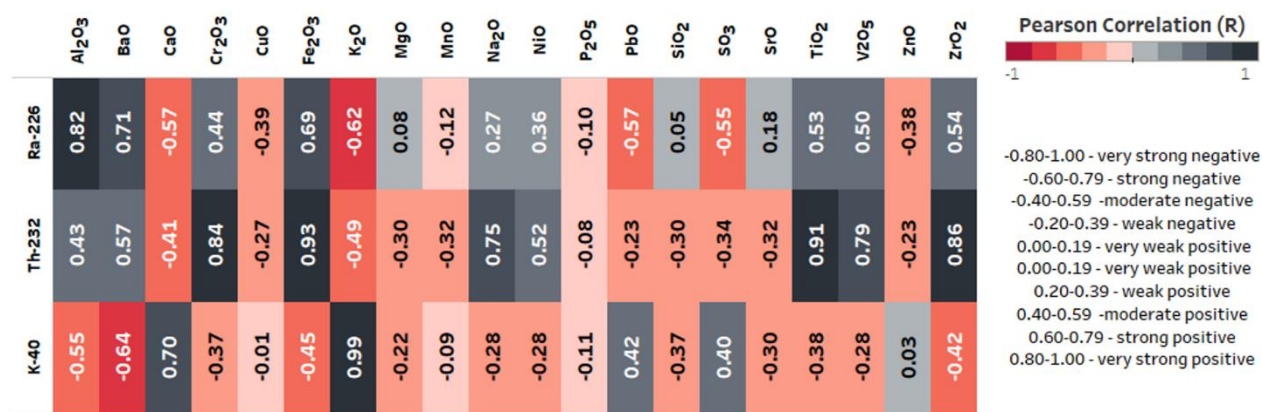
concentration. This can be explained by the condensation of these isotopes in the kiln during the cooling

368 phase of operation and their subsequent release during preheating when the CKD accumulates these
369 isotopes. However, the exact mechanism should be investigated in more detail in future research projects.
370 The APC, ISSA and RHA samples had also increased Pb-210 content with a factor of 6.7, 2.5 and 2.1,
371 respectively, which fits with the conclusion based on the volatility of Pb and Po. The FA samples did not
372 show significant Pb-210 accumulation which is not unusual in the case of FA residues. Relatively low
373 enrichment factors can also be found in the literature (Ozden et al., 2018).

374 **3.5 Correlation between NOR contents and chemical composition**

375 The correlation between the NOR isotopes and the chemical composition results is presented in Fig. 6. In
376 the figure, the dark colours indicate the strong correlation between the NOR isotopes and the chemical
377 composition data. From the results it can be concluded that the Ra-226 concentration has a strong positive
378 correlation with the Al₂O₃, BaO and Fe₂O₃ contents and strong negative correlation was found with the K₂O
379 content. In the case of Th-232, the Cr₂O₃, Fe₂O₃, Na₂O, TiO₂, V₂O₅ and ZrO₂ contents had strong positive
380 correlation with Th-232. The K-40 content showed strong correlation with the CaO and, of course, with the
381 K₂O content. Strong negative correlation was found when compared with the BaO content. No correlation
382 with the P₂O₅ content was observed for any of the NORs. Owing to the various industrial processes and the
383 results of the heat map, it can be concluded that the NOR isotope contents can be expected to correlate with
384 the oxides mentioned above.

385



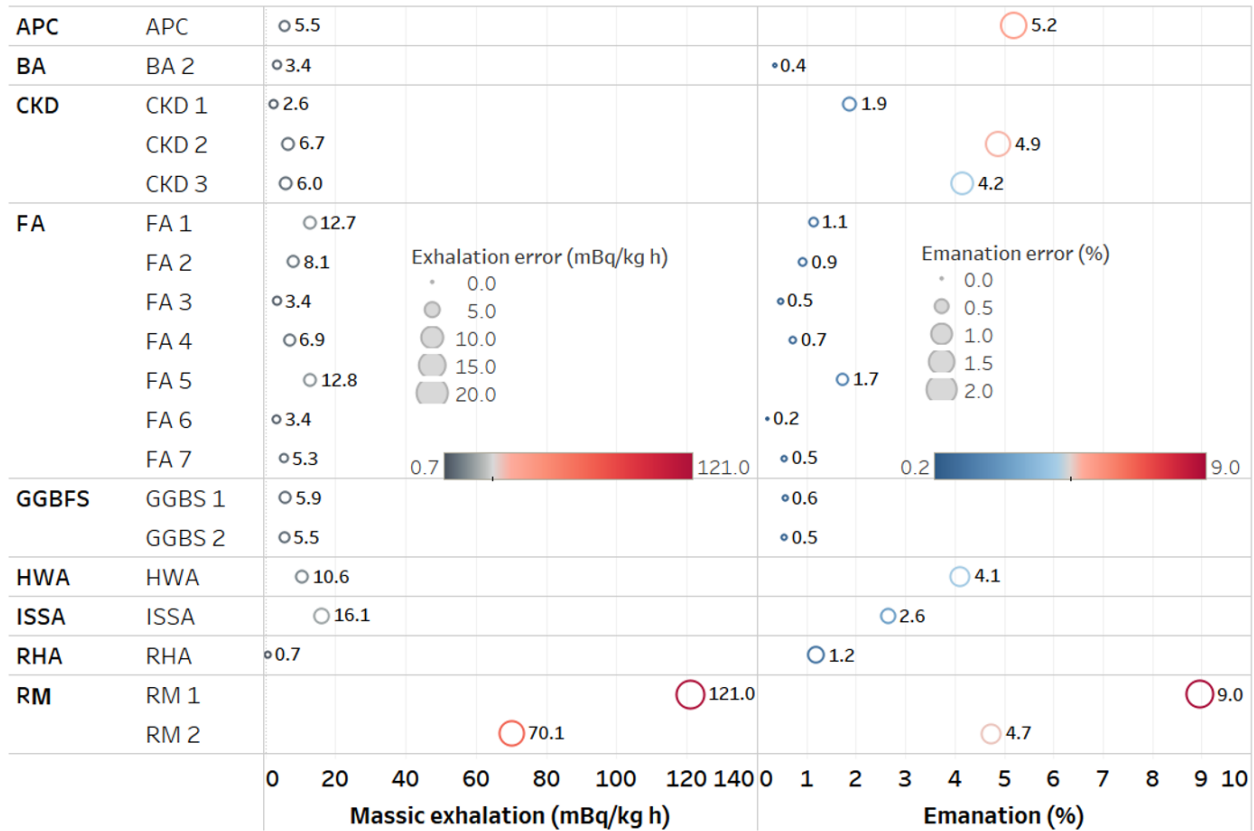
386
387 **Fig. 6:** Comparison between the naturally occurring radionuclide (NOR) isotopes and main chemical
388 compositions illustrated on a heat-map type diagram

389 The purpose of the correlation analysis was to examine the possibility of using the information about the
390 chemical composition as a source apportionment tool for NORs. This is useful considering the future
391 recovery of materials from old landfills, lagoons, or illegal dumps where information on the original material
392 is not available. Based on the Pearson correlation results, the strong correlation, either positive or negative,
393 shows that the concept is promising to predict the level of the expected NOR content. However, the
394 conclusion was drawn from 19 samples, so the extension of the dataset and further correlation analysis could
395 strengthen the concept.

396 3.6 Radon emanation and massic exhalation features of the studied samples

397 After the gamma spectrometry measurements, the massic radon exhalation of the samples was determined,
398 and their radon emanation factor was calculated with the obtained Ra-226 content. The obtained results are
399 illustrated in Fig. 7. The massic exhalation of the residues varied between 0.7 mBqkg⁻¹ h⁻¹ and 121.0 mBqkg⁻¹
400 h⁻¹ with an average of 16.1 mBqkg⁻¹ h⁻¹. The emanation factor was obtained between 0.6-8.6%. The average
401 of the emanation factor was 2.7%. This variation of the emanation factor of certain materials can be
402 explained by the different emanation coefficients, which depends on the characteristic of the material matrix
403 (Hegedus et al., 2016; Kovacs et al., 2016; Sas et al., 2015b). The results generally match those reported in
404 various publications (International Atomic Energy Agency, 2013). The two highest massic exhalation rates

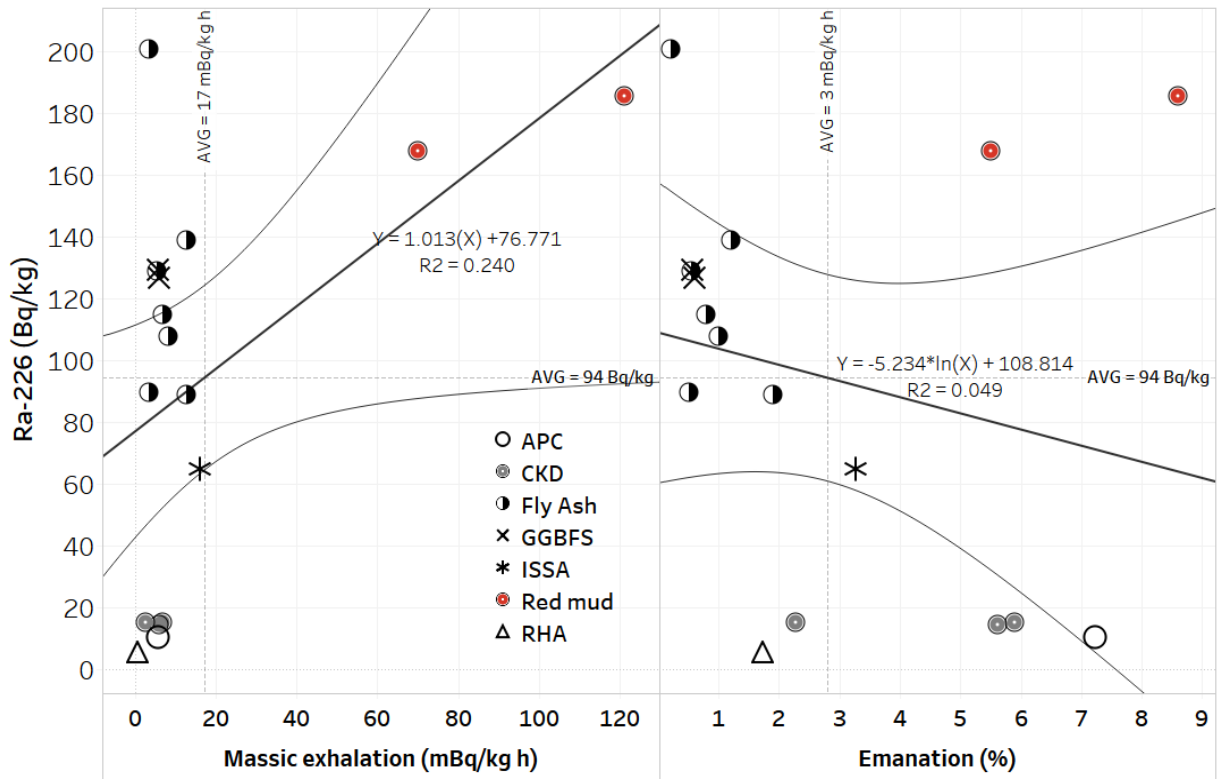
405 were found in the case of the red mud samples which can be explained by their significantly higher Ra-226
 406 content and their relatively high emanation coefficient. In the case of the fly ash samples, seven samples
 407 were analysed from different origins. The calculated emanation factors were under 2% in all the seven cases
 408 which fit well with the literature data (International Atomic Energy Agency, 2013; Sas et al., 2014).



409

410 **Fig. 7:** Massic exhalation and emanation factor of studied samples, where the error values were derived
 411 from the sensitivity of the instrument, using information provided by manufacturer

412 The massic radon exhalation depends on the Ra-226 content and the emanation factor of the matrix
 413 (International Atomic Energy Agency, 2013). That is why the belonging values and their correlation were
 414 also investigated. The results of the comparison are illustrated in Fig. 8.



415

416 **Fig. 8:** Correlation between the Ra-226 and massic exhalation results and Ra-226 and emanation factor
 417 with 95% confidence bands

418 However, during the analysis, the fitting of a linear trendline showed a weak correlation between the Ra-
 419 226 and the massic exhalation. The relatively low number of the samples does not make possible to state
 420 the relationship between these parameters explicitly. Furthermore, the emanation factor also influences the
 421 massic exhalation. Between the Ra-226 and the emanation factor no correlation was found which proves
 422 that they are independent parameters. Measuring the Ra-226 content by itself does not allow an assessment
 423 to verify whether the radon exhalation of building materials is increased or reduced.

424 4. Conclusions

425 From the results, there were no extraordinary NOR contents in the case of any of the samples. Compared
 426 with recently published NOR databases the values were close to the average values of the datasets. The

427 highest NOR contents were found in the case of the red mud samples which emphasize the importance of
428 the determination of its NOR contents before reusing it for construction material purposes.

429 The radioactive equilibrium state in the decay chains fits well with the scientifically reported data. In the
430 case of all the cement kiln dust, 13-16 times higher Pb-210 content compared to the Ra-226 activity
431 concentration was observed which requires further investigation to find out the exact reason for the
432 accumulation.

433 Significant positive correlation was found between the Ra-226 and the Al_2O_3 , BaO and Fe_2O_3 contents and
434 strong negative correlation were found with the K_2O content. Strong negative correlation was observed with
435 the CaO content. Th-232 had a strong positive correlation with the Cr_2O_3 , Fe_2O_3 , Na_2O , TiO_2 , V_2O_5 and
436 ZrO_2 content. The K-40 content had a strong positive correlation with the CaO and the K_2O content while
437 strong negative correlation was found with the BaO content.

438 The massic exhalation demonstrated a broad range. The highest values were found in the case of the red
439 mud samples. The emanation coefficients were significantly lower in the case of the residues generated as
440 a result of high-temperature processes. Compared with literature data (International Atomic Energy Agency,
441 2013), all the obtained emanation factors are regular.

442 These emanation coefficients, chemical correlations together with the radioactive equilibrium state could
443 be used to source apportion materials found during the process of landfill mining and recovery of material
444 for recycling, by, for example, mixing more active materials with less active materials to achieve an overall
445 acceptable mix from the radioactivity point of view.

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