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Direct Structural Identification and Quantification of the Split-Vacancy Configuration for Implanted Sn in Diamond

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We demonstrate formation of the ideal split-vacancy configuration of the Sn-vacancy center upon implantation into natural diamond. Using β^- emission channeling following low fluence ^{121}Sn implantation (2×10^{12} atoms/cm², 60 keV) at the ISOLDE facility at CERN, we directly identified and quantified the atomic configurations of the Sn-related centers. Our data show that the split-vacancy configuration is formed immediately upon implantation with a surprisingly high efficiency of $\approx 40\%$. Upon thermal annealing at 920°C $\approx 30\%$ of Sn is found in the ideal bond-center position. Photoluminescence revealed the characteristic SnV^- line at 621 nm, with an extraordinarily narrow ensemble linewidth (2.3 nm) of near-perfect Lorentzian shape. These findings further establish the SnV^- center as a promising candidate for single photon emission applications, since, in addition to exceptional optical properties, it also shows a remarkably simple structural formation mechanism.

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Point defects in diamond are being intensively investigated for their applications in quantum information science, as well as for quantum metrology [1]. So far, the negatively charged nitrogen-vacancy center (NV^-) has been the most studied solid-state qubit [1-6] thanks to its efficient optical spin polarization, spin-state dependent fluorescence, and long spin coherence time at room temperature (RT). Owing to their superior optical properties such as their higher Debye-Waller factor, the negatively charged group-IV-vacancy centers, mainly SiV^- [7-9] and GeV^- [10-11] are intensively studied, however, these centers have short spin coherence times. Lately SnV^- [12-20] and PbV^- [21-22] have emerged as highly interesting types of point defects for diamond-based quantum nanophotonics and quantum communication, due to expected longer coherence times and this at higher temperatures than SiV^- , as was recently reviewed in Refs. [23-24]. Nonetheless many key properties and especially the formation mechanism and local configuration of these defects are largely unexplored.

It is generally accepted that, whereas the N atom in the NV^- center occupies a substitutional carbon site, the group-IV-vacancy centers form a so-called split-vacancy configuration, where the group IV impurity atom sits in between two adjacent C vacancies in a position that would correspond to a bond-centered (BC) site in an unperturbed lattice (Fig. 1). This configuration was first proposed for the Si impurity by Goss *et al.* [25-27] based on *ab-initio* local density calculations, and later also extended by theory to the other group IV elements Ge, Sn, and Pb [22,26,28-29].

Experimentally, however, the group-IV-vacancy structural configurations have only been indirectly supported, e.g., from the number of optical emission lines and their polarization and magnetic field splitting [19-20,30-31]. Techniques for direct, detailed, and quantitative characterization of the structure of these defects are especially important for the

quantum technology field, since the superior optical properties of the group-IV-vacancy centers are to a large extent a consequence of their D_{3d} inversion symmetry rather than C_{3v} , as in the case of NV^- . For a split-vacancy configuration, however, D_{3d} symmetry will only be realized if the impurity is exactly centered within the divacancy, which corresponds to the ideal BC position. Currently the most used technique for creating group-IV-vacancy centers is ion implantation [7-14,17-24,31-32]. The fractions of implanted impurities that form the relevant *optically active* color centers can then be obtained from comparing the number of centers per area found with confocal luminescence microscopy to the implanted fluence. For undoped diamond, this yielded formation efficiencies of optically active SnV^- centers around 5% [12], 1%-2% [13], 0.7% [17], and 2.5% [18] of implanted atoms only. By carefully engineering Sn implantations into phosphorus-doped diamond, the highest optical activation for SnV^- reported so far [18] is 8.6% after annealing at 1200°C. Because of the lack of structural characterization methods it is unknown whether higher activation yields are inhibited because the relevant SnV^- centers are not formed or because they are formed but in an optically inactive state or their optical activity is quenched.

In this letter, we provide direct lattice location measurements for implanted radioactive ^{121}Sn in diamond using the β^- emission channeling (EC) technique. EC is uniquely suited to characterize and quantify the possible point defect configurations via the lattice sites of the impurity atoms. In particular, this structural technique is able to work at the low implantation fluences ($\approx 10^{12}$ cm⁻²) that are often used to produce quantum color centers in diamond. We show that a surprisingly large fraction of Sn ($\approx 40\%$) already forms SnV^- centers directly upon implantation without any annealing. Additionally, we confirm formation of SnV^- optically by confocal photoluminescence (PL) resulting from long-lived

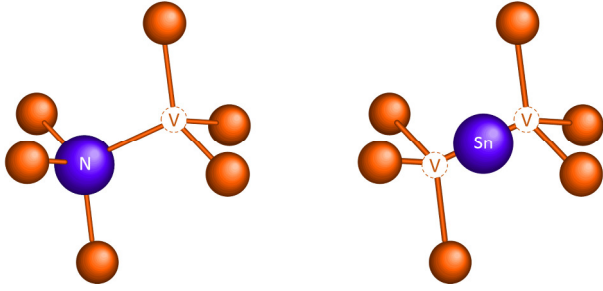


FIG. 1. The two major structures that are generally considered for impurity-vacancy complexes in diamond. It is assumed that with a vacancy V next to it the N impurity remains in a substitutional site (left), while for Sn the so-called split-vacancy configuration (right) has been proposed [25, 28]: the Sn atom moves from its substitutional site to a position midway between the two adjacent vacancies, which geometrically coincides with the so-called bond-center position in an undisturbed lattice.

^{121m}Sn in the same sample. Following annealing at 920°C , both EC and PL find SnV centers in a single, well-defined atomic configuration, leading to a remarkably narrow luminescence linewidth.

The basic concept of the β^- EC technique is that the lattice sites of implanted radioactive isotopes in single crystals are probed via the channeling of emitted beta decay particles [33-35]. The underlying physical effect is the guidance of β^- particles by the crystal potential, which strongly depends on the lattice site from which the particle originated. This leads to an angular dependence of the β^- emission yield in the vicinity of major crystallographic directions, which is recorded by a two-dimensional position-sensitive detector [36], thus providing patterns that are characteristic for the probe atom lattice location in the sample. The fact that EC is based on the detection of decay particles from radioactive probes makes it a factor of $\approx 10^4$ more efficient [35] than conventional lattice location techniques using external ion beams. The capability of EC to identify impurities on the BC site in a diamond lattice structure following implantations at the 10^{12} - 10^{13} cm^{-2} level has been previously demonstrated, e.g., in Ge for ^{121}Sn [37] and for the transition metals ^{59}Fe , ^{67}Cu and ^{111}Ag [38].

For determining the lattice location of Sn, we have used the short-lived radioactive isotope ^{121}Sn , which decays with a half-life of $t_{1/2}=27.06$ h via β^- emission (β^- endpoint energy 400 keV, average 116 keV) into stable ^{121}Sb , while the detected PL signal from SnV^- resulted from the metastable, long-lived isomeric state ^{121m}Sn ($t_{1/2}=55$ y). Both ^{121}Sn in its ground and isomeric state were produced and, due to their near identical mass, simultaneously implanted at CERN's ISOLDE on-line isotope separator facility [39-41]. The sample was a natural type IIa diamond (Dutch Diamond Technologies, 3×3 mm^2 size, thickness 300 μm), with the surface cut and polished 8° from a $\langle 110 \rangle$ direction. The implanted fluence of 60 keV $^{121}\text{Sn}+^{121m}\text{Sn}$ was 2.3×10^{12} atoms/ cm^2 , into a beam spot of 1 mm diam (average current 3 pA). According to SRIM simulations [42], this results in a projected range and straggling of 211 ± 40 \AA , Sn peak concentration 2.3×10^{18} atoms/ cm^3 , mean Sn-Sn distances larger

than 76 \AA , and each implanted atom initially creating ≈ 430 lattice vacancies. By obtaining the number of implanted ^{121}Sn atoms from the measured β^- count rate, the ratio of ^{121}Sn : ^{121m}Sn was estimated to be $\approx 44:56$ [43].

In order to derive the fractions of ^{121}Sn on different lattice sites, the angular-dependent experimental β^- emission yields were fitted by theoretical patterns corresponding to ^{121}Sn emitter atoms on up to two different lattice sites. The theoretical β^- patterns were calculated by means of the so-called “many-beam” formalism outlined in Refs. [33-34,43-47]; general aspects of the fit procedure are described in Refs. [34-36,48]. The major lattice sites in the diamond structure that were considered in the simulations are shown in the supplemental material [43]. They included substitutional (S) and ≈ 250 interstitial sites which are obtained by displacing from the S position along $\langle 110 \rangle$, $\langle 100 \rangle$, or $\langle 110 \rangle$ directions in steps of ≈ 0.04 \AA . Besides such static displacements, we also calculated the expected yields for sites that are spread around S and BC positions with Gaussian probability distributions of larger root mean square (rms) than expected from thermal vibrations alone, which for a diamond Debye temperature of 1860 K [49] is $u_1(^{121}\text{Sn})=0.0337$ \AA [43].

Figures 2(a)-2(d) show the angular distribution of β^- particles emitted around the $\langle 110 \rangle$, $\langle 211 \rangle$, $\langle 100 \rangle$ and $\langle 111 \rangle$ directions in the as-implanted state, while Figs 3(a)-3(d) are following 920°C annealing for 10 min in vacuum. The fact that all axial and planar directions exhibit channeling effects shows that a large fraction of emitter atoms occupies sites close to the substitutional positions. Note that the anisotropy of the channeling effects roughly doubles upon annealing, which is a clear indication of damage annealing and/or a change of the configuration of Sn in the lattice.

The experimental patterns were first fitted using one-site fits, i.e., allowing for only one of the calculated positions in the lattice to be occupied. A detailed description of the analysis procedure can be found in the supplemental material [43]. However, neither in the as-implanted nor in the annealed state simple ideal substitutional sites provided the best one-site fits; these were rather obtained for sites that are displaced from the ideal S sites: in the as-implanted state displacements of 30%-50% (0.23-0.39 \AA) of the distance from S to BC sites yielded chi square (χ^2) improvements up to 44% (average for all four patterns 24%) compared to ideal S sites; following 920°C annealing displacements of 10%-30% (0.08-0.23 \AA) improved χ^2 by up to 19% (average 9%). However, χ^2 further improved by allowing Sn on two different lattice sites in the fit procedure. In the as-implanted state the overall best fit was obtained for a combination of sites which are spread around the S and BC positions with rms displacements of $u_1(^{121}\text{Sn})$ of 0.18 and 0.11 \AA , respectively. While this combination of sites led to improvements up to 45% (average 26%) compared to ideal S sites, this result was only slightly better than when one statically displaced site was considered; it can therefore not be discerned whether in the as-implanted state a single site located in between S and BC or a combination of S and BC sites with larger rms amplitudes prevails. In contrast, in the

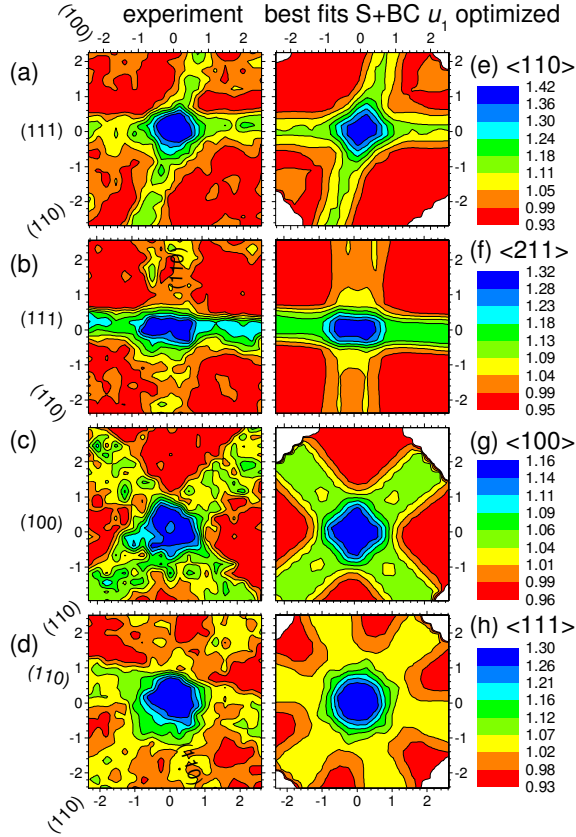


FIG. 2. (a)-(d): Experimental β^- emission channeling patterns from ^{121}Sn in diamond around the $\langle 110 \rangle$, $\langle 211 \rangle$, $\langle 100 \rangle$, and $\langle 111 \rangle$ directions in the as-implanted state. The plots (e)-(h) are the corresponding best fits of theoretical patterns considering 63% on substitutional and 41% on bond-center sites with rms $u_i(^{121}\text{Sn})$ of 0.18 Å and 0.11 Å, respectively. Note that due to the unusual surface cut of the diamond, it could not be oriented in all cases in such a way that the calculated patterns [$\pm 3^\circ$ with (110) plane horizontal] cover the whole range of measured angles. The area in the best-fit patterns (e)-(h) which is not covered is shown in white.

920°C annealed state the best fit was clearly obtained for a combination of ideal S and BC sites, resulting in significant χ^2 improvements (up to 24%, average 12%) compared to S sites only. The theoretical patterns that correspond to the best fits are shown in Figs 2(e)-2(h) and 3(e)-3(h) and match the experimental results in considerable detail.

For the analysis of the as-implanted state we adopted the interpretation of two occupied sites; in this case (63±6)% of ^{121}Sn is assigned to S and (41±5)% to BC sites, however, as was already mentioned, both sites are not the ideal ones but subject to Gaussian spreads. Using a single site located in between S and BC would result in a lower fraction of 81% only. The situation following 920°C annealing is characterized by two major changes: (a) BC as well as S sites become ideal, (b) fractions are now $f_S=(79\pm 6)\%$ and $f_{BC}=(32\pm 5)\%$, i.e. some of the ^{121}Sn on BC has been converted to S sites. Both effects together are responsible for the pronounced change in the anisotropy of the channeling patterns upon annealing. The sum $f_{\text{sum}}=f_S+f_{BC}$ of the two fitted fractions reaches values slightly above 100%, which is related to an

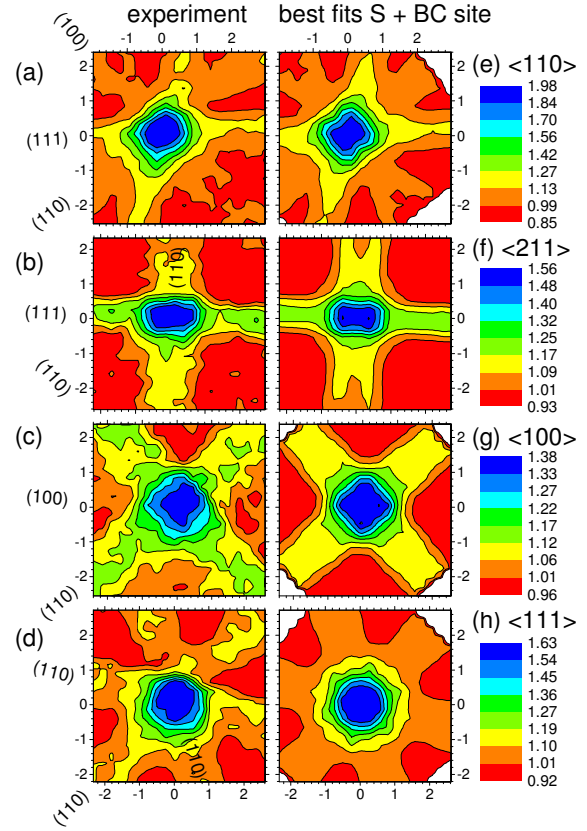


FIG. 3. (a)-(d): Experimental β^- emission channeling patterns from ^{121}Sn following 920°C annealing. The plots (e)-(h) are the corresponding best fits of theoretical patterns considering 79% on ideal substitutional and 32% on ideal bond-center sites.

overestimation of the background correction from scattered electrons, as is explained in the supplemental material [43].

Confocal PL spectra (Fig. 4) following annealing at 920°C show the prominent first order Raman peak at 573 nm and also the weaker second order Raman line at 612 nm. Focusing below the surface, the characteristic features of the NV^0 zero phonon line (ZPL) at 575 nm and the NV^- ZPL at 636 nm and associated phonon side bands at longer wavelengths are visible, reflecting the presence of nitrogen-associated defects in the natural IIa diamond. Detailed confocal scanning of the sample [43] showed that in the spots where low NV signals are found, close to the implanted surface a strong, sharp peak is detected at 621.3 nm, corresponding to the well-known ZPL of SnV^- [12-21,23-24,29]. Fitting of this line using a Lorentzian [43] shows a full width half maximum (FWHM) of 2.3 nm, the narrowest RT linewidth reported to date for the ZPL of an ensemble of SnV^- [12-16,19], cf. Ref. [43] for an overview on linewidths. The near ideal Lorentzian shape [43] implies a very narrow distribution of single center emission lines in the whole ensemble. The 621-nm line is accompanied by much weaker peaks at 631 and 643 nm, in good agreement with Refs. [12-14,18]. The peaks at 621, 631, and 643 nm weaken as the focus is set deeper into the sample, away from the implanted surface. A sharp PL line at 581.4 nm (FWHM 1.5

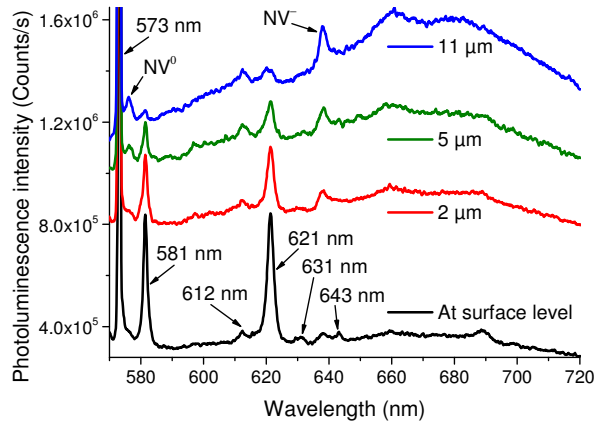


FIG. 4. Confocal photoluminescence spectra from the diamond sample implanted with $^{121}\text{Sn}+^{121\text{m}}\text{Sn}$ ions after annealing at 920°C and following the decay of ^{121}Sn to ^{121}Sb , i.e. SnV related lines result from long-lived $^{121\text{m}}\text{Sn}$. The spectra were recorded at room temperature under 532 nm laser excitation with the focus at various depths. Note that the depth resolution is $\approx 4 \mu\text{m}$ due to the Rayleigh range of the used objective with numerical aperture 0.95.

nm) is additionally observed close to the surface. This line could correspond to the L1 center, a bright defect of unknown nature presenting a narrow ZPL at 581.7 nm (reported FWHM <1.4 nm), which has been detected in diamond implanted with a mixture of molecular ions of mass 27 and annealed at 800°C [32,50]. For a discussion of the possible sources for this line we refer to the supplemental material [43].

From the combined results of EC and PL, we obtain unique insight in the structure and formation mechanism of SnV centers. First of all we have provided direct evidence that considerable amounts of Sn atoms implanted in diamond are found on lattice sites that correspond to the split-vacancy configuration of the SnV center. The microscopic model so far assumed for SnV centers and their D_{3d} symmetry is thus fully supported by our results, obtained with a technique that directly probes structure.

Second, our EC results show that the fraction of implanted Sn on BC sites is surprisingly high, about an order of magnitude above the formation yield of optically active SnV centers reported from confocal PL [12-13,17-18]. If these previous studies should have achieved the same structural formation yield of BC Sn, this would mean that $\approx 90\%$ of the SnV centers were optically inactive at 621 nm, e.g. because they exist in a different charge state than SnV, such as SnV⁰, do not possess the right D_{3d} symmetry, or are prevented from being optically active by some yet unknown mechanism, e.g., H passivation [51]. In this respect we note that all previous Sn implantation studies (cf. list in Ref. [43]) have used synthetic diamond grown by chemical vapor deposition (CVD), which tends to be rich in H, while we used a natural diamond. By pre-doping CVD diamond with the deep donor phosphorus prior to Sn implantation, Lühmann *et al.* recently achieved optical activation of SnV in the 8%-9% range [18]. This was attributed to vacancies in

diamond becoming negatively charged V_C^- in the presence of donors, thus being prevented from agglomerating by Coulomb repulsion and leading to “a more efficient formation of SnV together with a reduced formation of competing centres” [18]. Since EC can quantify the presence of ^{121}Sn on BC sites from the measurement of a single $\langle 110 \rangle$, $\langle 211 \rangle$ or $\langle 100 \rangle$ emission pattern, which requires an implanted fluence around $1 \times 10^{11} \text{ cm}^{-2}$ only, it seems possible to test in future experiments whether P doping directly changes the structural formation efficiency of SnV, as suggested in Ref. [18], or rather favors its activation as an optical center.

Third, our studies shed light on the creation of SnV centers during ion implantation and the role of thermal annealing. The centers are clearly already formed in large quantities during the implantation process and the subsequent cool down of the thermal spike caused by the collision cascade, although in a perturbed configuration that is attributed to nearby radiation damage such as vacancies and interstitials. Considering the formation and binding energies for Sn defects in diamond calculated by Goss *et al.* [26], from a thermodynamic point of view this can be understood as follows, using similar arguments as presented for Sn in Ge [38]. Isolated substitutional Sn is predicted to have the lowest heat of formation of all Sn defects and should hence be the dominant species; however, when it encounters an additional vacancy it becomes energetically favorable to form SnV and change its lattice site to the BC position. The energy gain associated with this process is predicted in the 8-9 eV range [26]. While the required vacancies are abundantly available and also mobile during the cool-down of the thermal spike following implantation of each ion, SnV formation has to compete with the annihilation of vacancies by carbon interstitials. As our results show, thermal annealing is not required in order to bring additional vacancies towards the implanted Sn (as is sometimes wrongly assumed in analogy to NV). The annealing process serves two purposes: (a) restoring the optical properties of diamond after the implantation, (b) restructuring the SnV center into its ideal configuration where Sn resides on *unperturbed* BC sites. To these effects, diffusion of vacancies is still essential for *removing* defects, in particular also from the immediate SnV neighborhood, which has recently been studied using a theoretical approach [52]. It was predicted that V_C -SnV complexes are thermodynamically more stable than their components, i.e., isolated SnV and vacancies V_C , which are both present in implanted samples. The formation of such complexes may therefore be responsible for the larger rms displacements of Sn on BC sites that we observed in the as-implanted state of our sample. The activation energy for breaking up V_C -SnV complexes was calculated as $E_A=2.4-3.4$ eV (depending on the charge state), with an attempt frequencies of $\nu_0 \approx 10^{14} \text{ s}^{-1}$. Using simple Arrhenius reaction rates one estimates that the expected lifetime $\tau = 1/[\nu_0 \exp(-E_A/k_B T)]$ of the complexes at 920°C is only 0.14 ms to 2.3 s, before the V_C diffuses away, leaving behind undisturbed SnV.

Fourth, the PL characterization performed on the 920°C annealed sample revealed ensembles of $^{121\text{m}}\text{SnV}$ ZPLs with near-perfect Lorentzian shape and with the narrowest RT ensemble linewidth reported so far (2.3 nm, just slightly larger than a recently found single center linewidth of 1.98 nm [20]). It is thus clearly possible to produce entire ensembles of ion implanted SnV centers with a narrow spread of configurations without the need for annealing much higher than 900°C. However, the exact conditions that are required to achieve ensembles of optically active SnV centers with such a narrow linewidth are still unclear, although low implantation fluences seem to be crucial [43]. While in our natural diamond sample optically active SnV centers were only detected in areas with low PL signals from NV centers, whenever an ensemble of SnV lines was found, it exhibited a narrow linewidth [43].

Last but not least, the EC technique has been found well suited to investigate the structural nature of ensembles of implanted optical centers in diamond. Suitable EC probes for further studies of group-IV-vacancy centers could be ^{31}Si ($t_{1/2}=157$ min), ^{75}Ge (82.8 min), and ^{209}Pb (3.2 h); for other elements of interest as single photon emitters also suitable β^- emitters exist, e.g., ^{27}Mg , ^{45}Ca , ^{89}Sr , ^{65}Ni , or the noble gases ^6He , ^{23}Ne , ^{41}Ar , ^{87}Kr , and ^{135}Xe . In many cases EC could be ideally complemented by radiotracer PL, which allows the direct chemical identification of PL lines via correlating their change in intensity with the half-life of radioactive decay [40].

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