

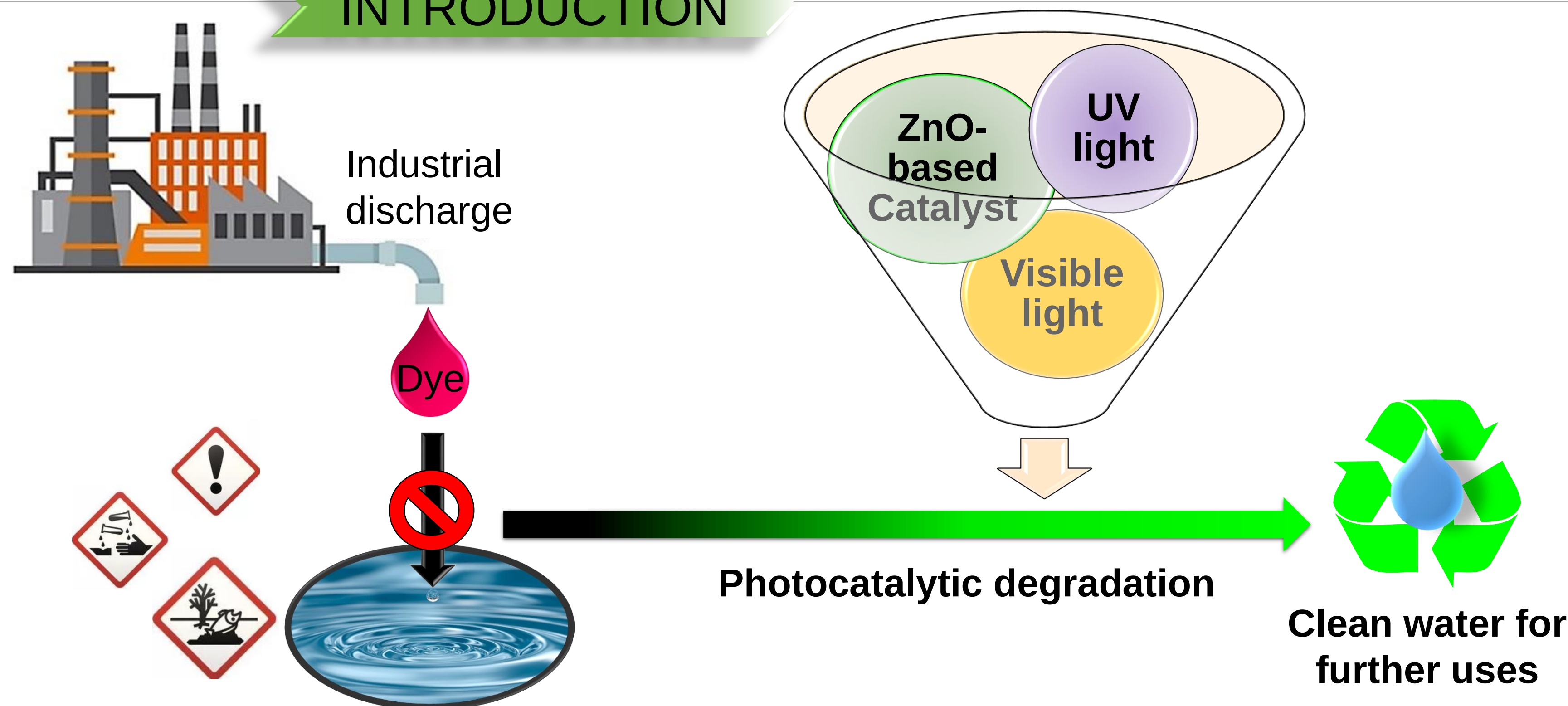
Undoped and Al-doped ZnO nanoparticles for photocatalytic application

Alessandra Piras^{1,2}, Ken Elen², Luca Fusaro¹, Peter Adriaensens², An Hardy², Carmela Aprile¹, Marlies Van Bael²

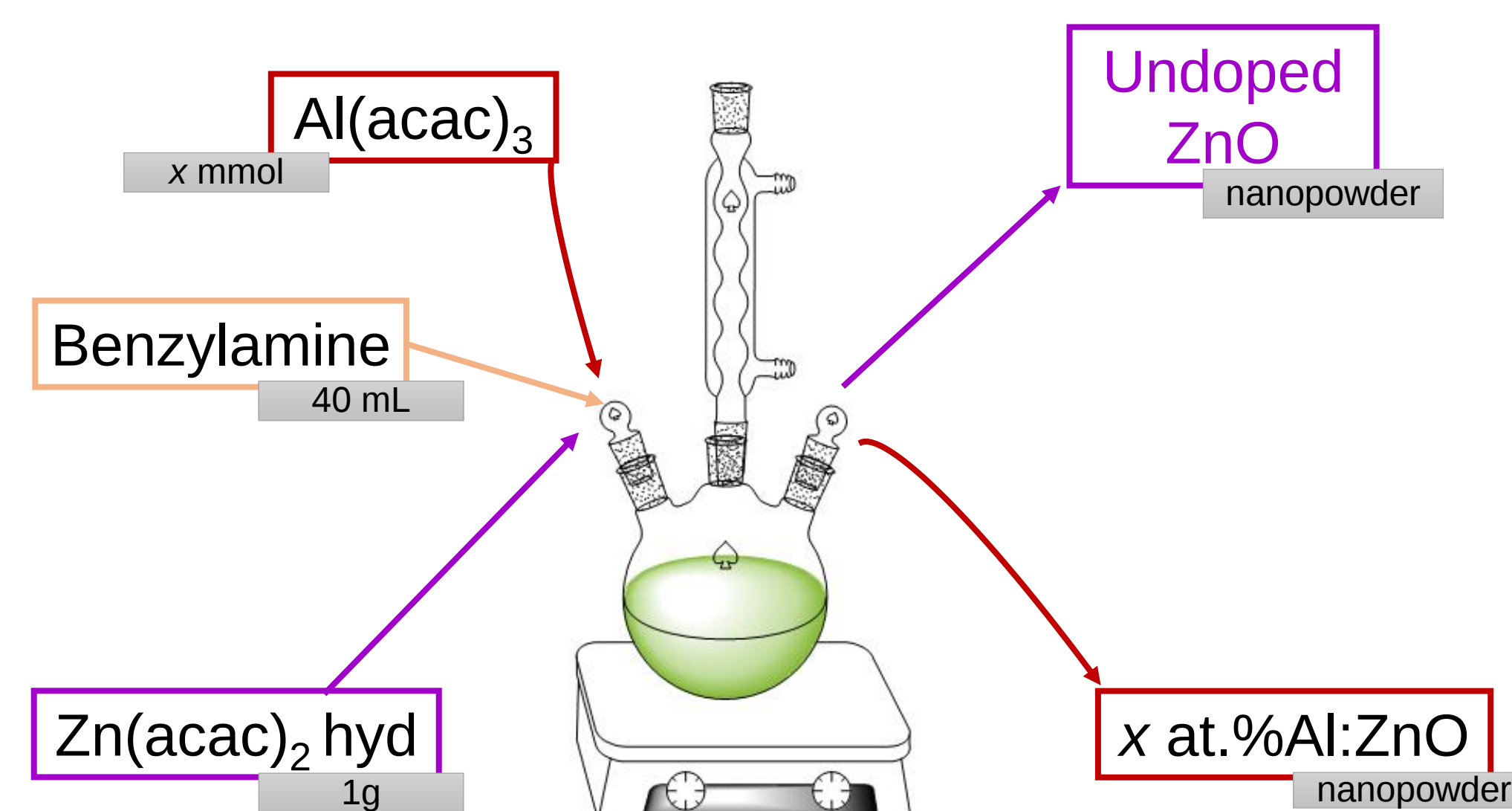
1. Namur University, Chemistry department, Applied Materials Chemistry (CMA), Rue de Bruxelles 61, 5000 Namur, Belgium

2. Hasselt University, Institute for Materials Research, DESiNe group, and imec, division imomec, Campus Diepenbeek, Agoralaan building D, 3590 Diepenbeek, Belgium

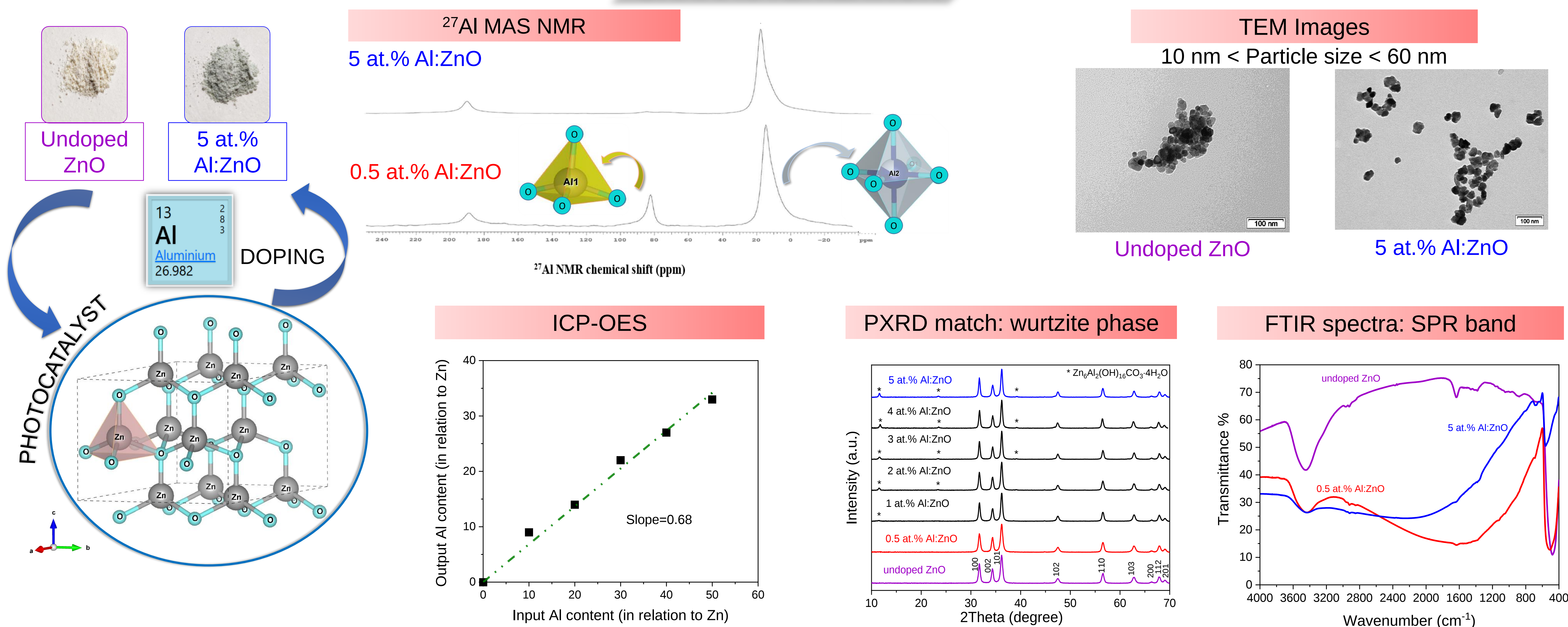
INTRODUCTION



SYNTHESIS

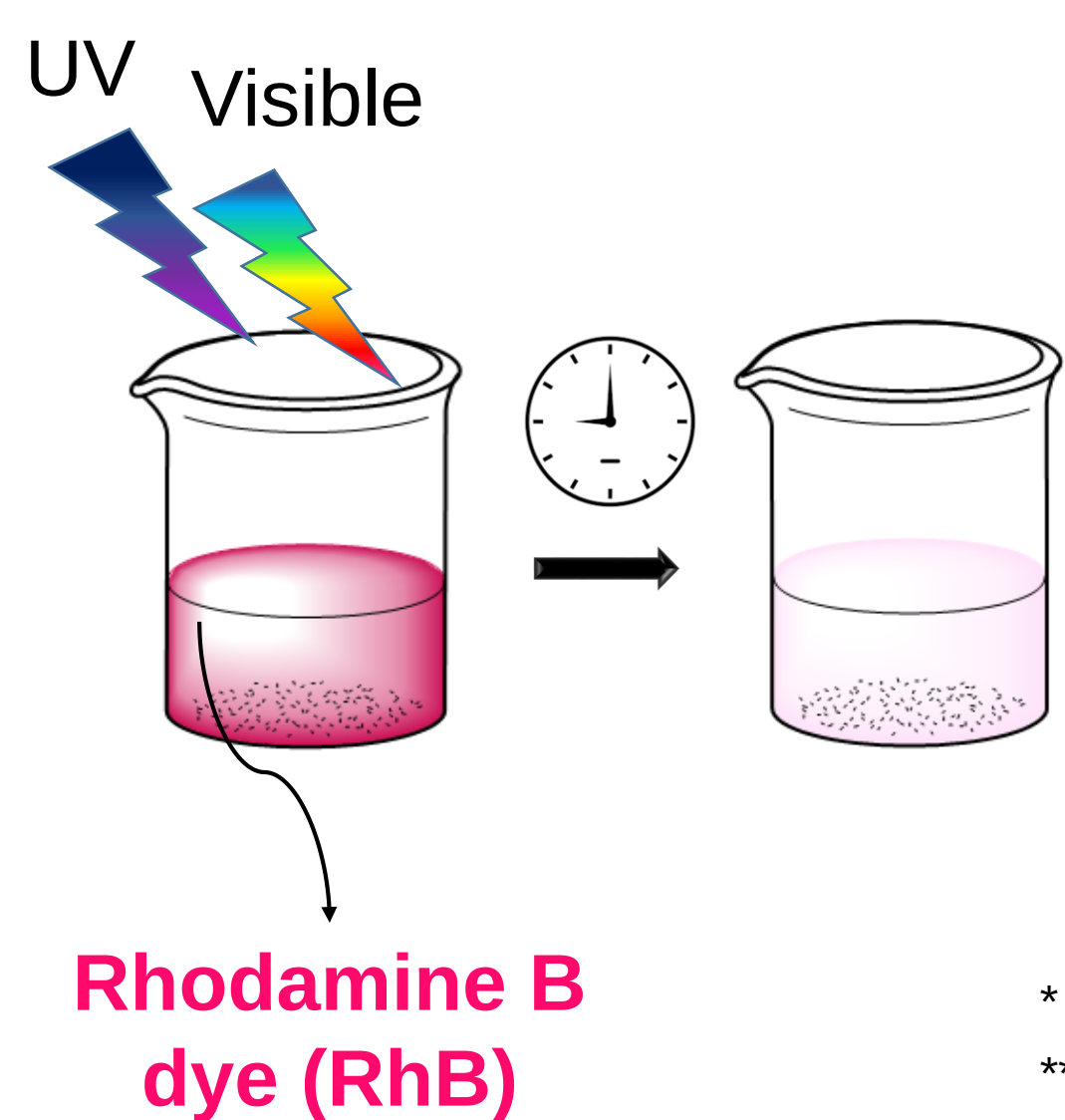


CHARACTERIZATION



CATALYTIC RESULTS

Photocatalytic degradation

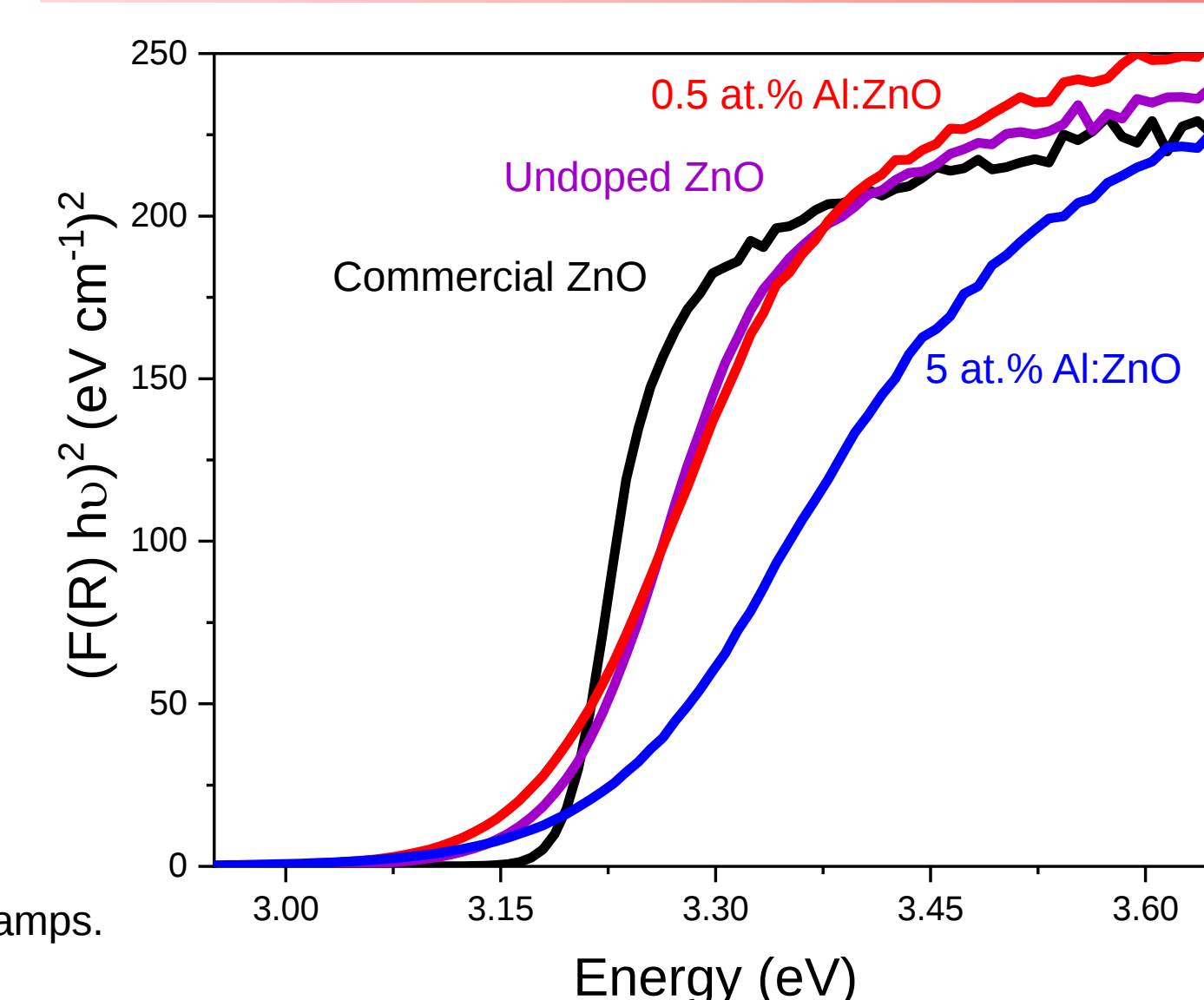


Material type	RhB Dye	RhB Dye	RhB Dye
	Degradation [%]**	Degradation [%]	Degradation [%]
	UV-C 254 nm*	UV-A 368 nm*	Visible 450 nm*
Commercial ZnO	94	96	8
Undoped ZnO	45	70	20
0.5 at.% Al:ZnO	41	53	22
5 at.% Al:ZnO	21	19	12

* Reactions performed at RT, 5 mg of catalyst, 20 mL of dye (RhB, 4 mg/L) for 20 min using 11W OSRAM UV-C, 11W OSRAM UV-A and 30W Strip LED lamps.

** Degradation % = $\frac{(A_0 - A)}{A_0} \times 100$, A_0 is the Absorbance of the starting solution before reaction and A is the final absorbance measured after reaction.

Energy band gap



CONCLUSIONS

Synthesis of undoped ZnO and Al:ZnO materials (0-5 at.%)

Physical - chemical characterization

Photodegradation: undoped ZnO and 0.5 at.% Al:ZnO showed the best performance under visible light

Correlation between Al positions and photocatalytic activity

Near future plan

Solar light simulator for photocatalytic dye degradation

Future plan

References

1. I. John Peter, E. Praveen, G. Vignesh and P. Nithiananthi, *Mater. Res. Express*, 2017, **4**, 124003.
2. A. Momot, M. N. Amini, G. Reekmans, D. Lamoen, B. Partoens, D. R. Slocombe, K. Elen, P. Adriaensens, A. Hardy and M. K. Van Bael, *Phys. Chem. Chem. Phys.*, 2017, **19**, 27866–27877.