

Unprecedented xenon collection and separation from air on silver-exchanged zeolites

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Belgian Nuclear Research Centre



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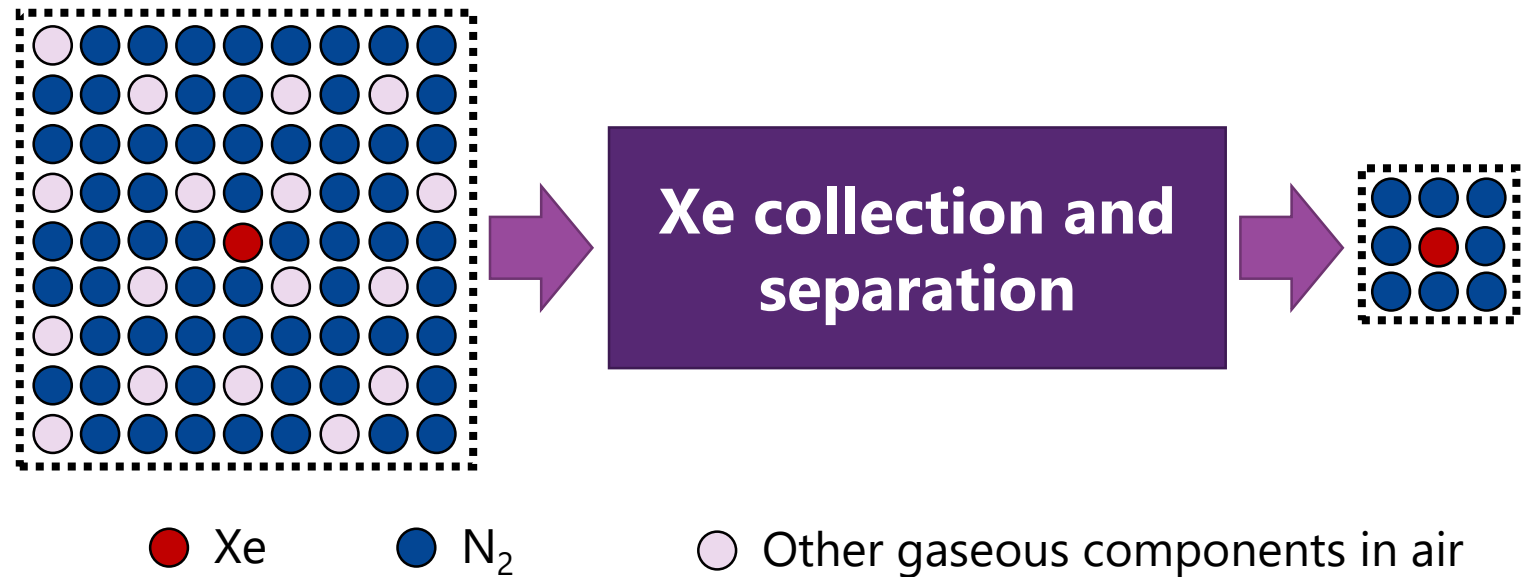
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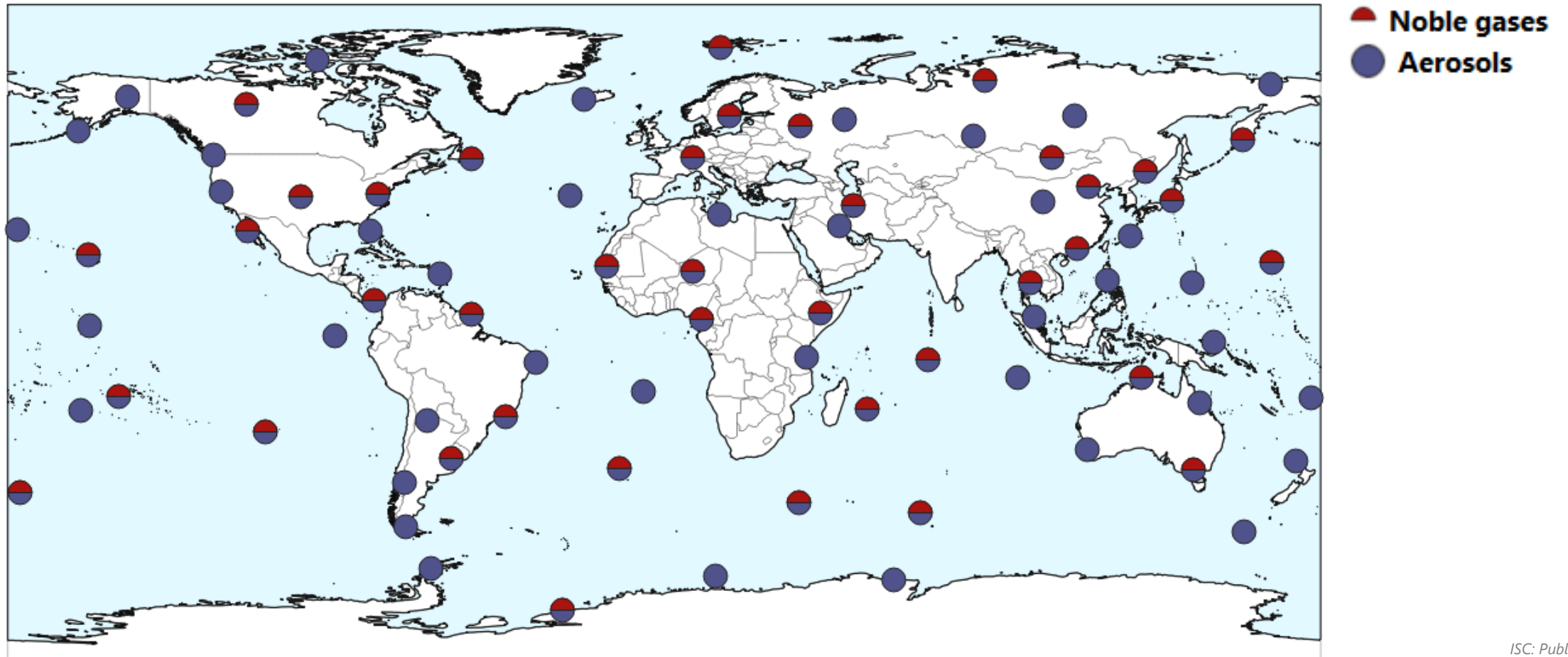
Content

- Introduction
- Objectives
- Xe collection
- Xe separation
- Conclusions & perspectives



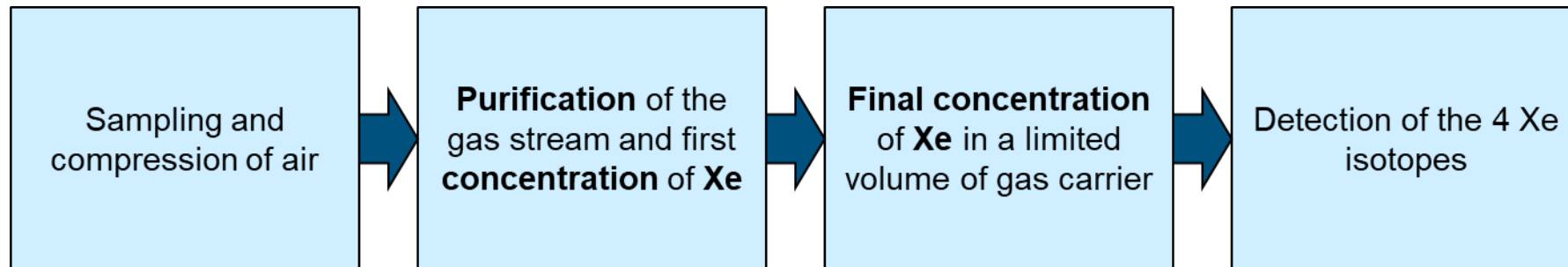
Introduction

- Radioxenon in the atmosphere is continuously monitored in the International Monitoring System for the verification of the CTBT



Introduction

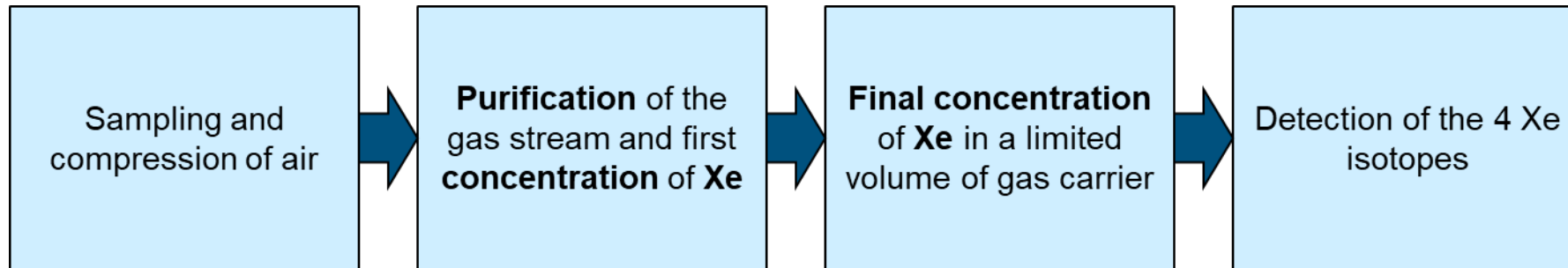
- Noble gas (radioxenon) monitoring systems
 - Requirements
 - 1 mBq/m³ (Xe-133), 10 m³ of air, 24h collection/measurement time, 95% data availability
 - 4 stages: air sampling, separation, concentration and detection
 - Quite complex and energy demanding separation systems



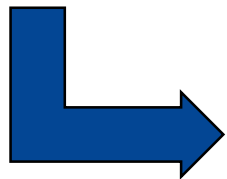
- < 2020
 - 2nd stage: remove H₂O and CO₂ using zeolites/molecular sieves/permeation membranes
 - 3rd stage: concentrate Xe on activated carbon (AC) at 100-150 K or room temperature

Introduction

- Noble gas (radioxenon) monitoring systems
 - 4 stages: air sampling, separation, concentration and detection



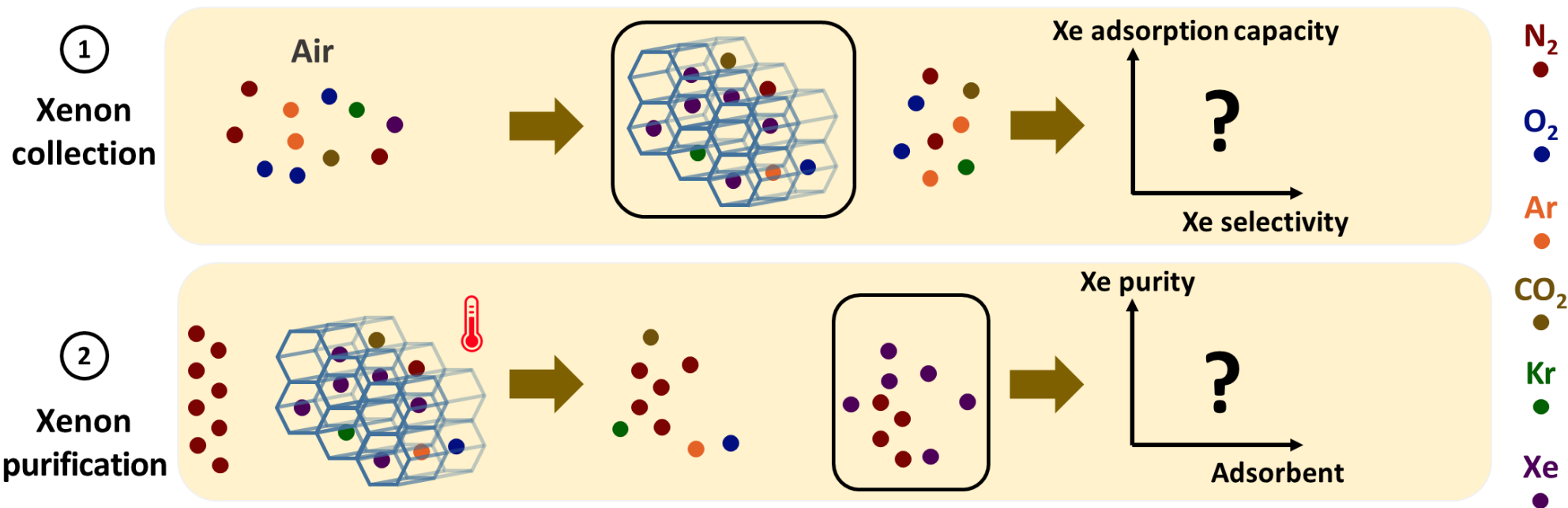
- Some new systems > 2020
 - 3rd stage: silver-exchanged zeolites (AgZs) at room temperature
 - Decreased energy consumption during desorption (smaller columns)



AgZs both for the 2nd and 3rd stage ?
Metal-Organic Frameworks (MOFs) ?

Objectives

1. How efficient and selective are AgZs and MOFs, compared to AC, in collecting Xe from atmospheric air ?
2. How easy and in which purity can we thermally recover the collected Xe from AgZs and MOFs compared to AC ?



Selected adsorbents

- Reference adsorbent
 - Activated carbon: Nusorb GXK
- Silver-exchanged zeolites used in new systems
 - Ag-ETS-10 & Ag-ZSM-5
- MOFs – Selection criteria (end 2018)
 - Only commercially available MOFs
 - > 10 g
 - Hydrothermal stability
 - Xe adsorption properties (from Xe/Kr separation literature)

Activated Carbon (AC)



Nusorb GXK

Silver-exchanged Zeolite (AgZ)



Ag-ETS-10
(SAUNA III)



Ag-ZSM-5
(SPALAX-NG)

Metal-Organic Framework (MOF)



Ni-DOBDC



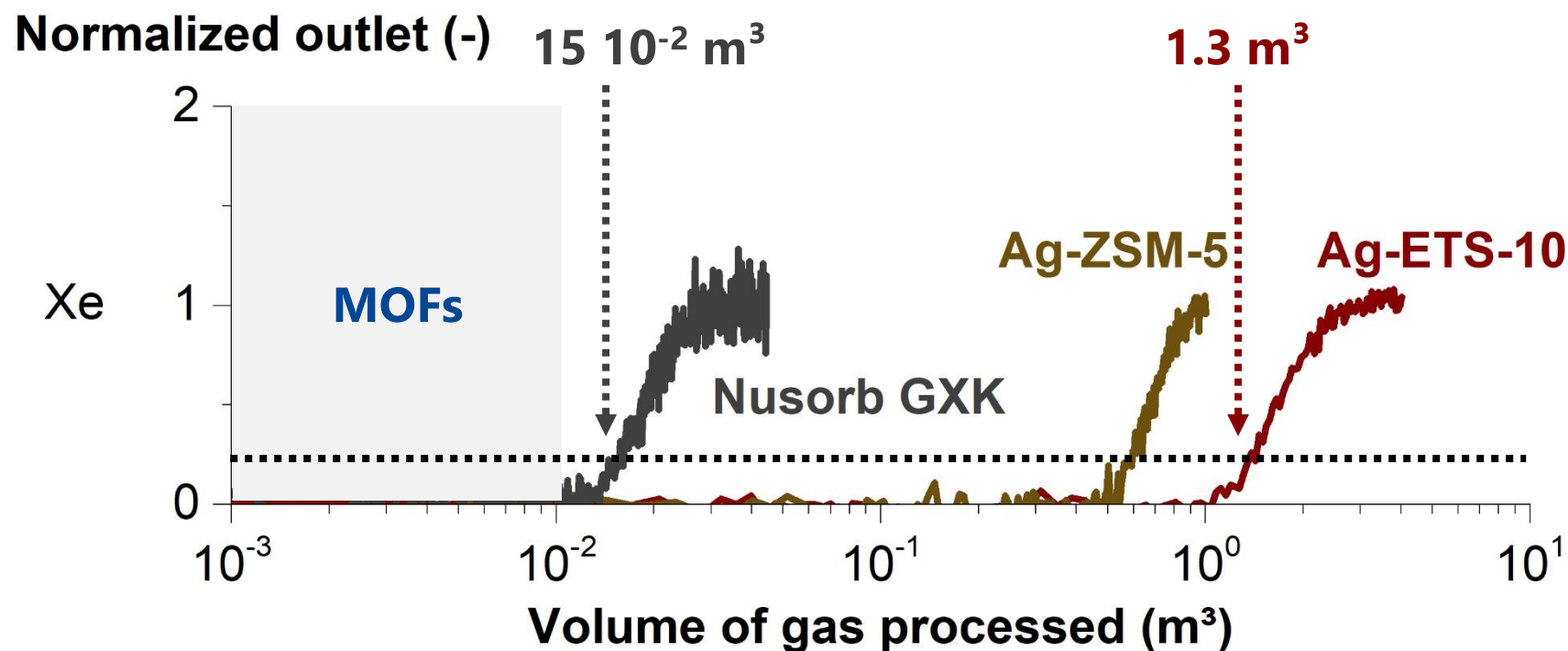
HKUST-1

1

How efficient and selective are AgZs and MOFs, compared to AC, in collecting Xe from atmospheric air ?

Xe collection from air at room temperature

- Breakthrough of Xe in dry air (87 ppb) at 0.12 m³/h in 50 cm³ of adsorbent

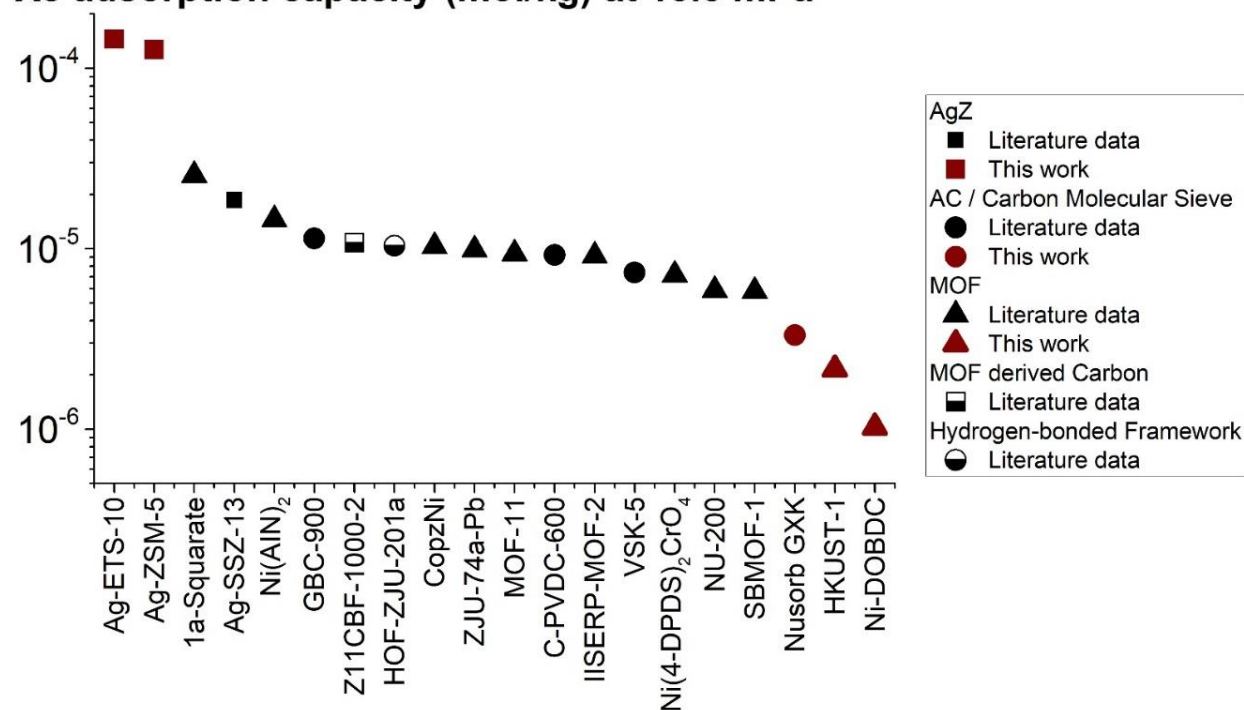


+ Xe selectivity over air on Ag-ETS-10 about 60 times higher than AC (> MOFs)

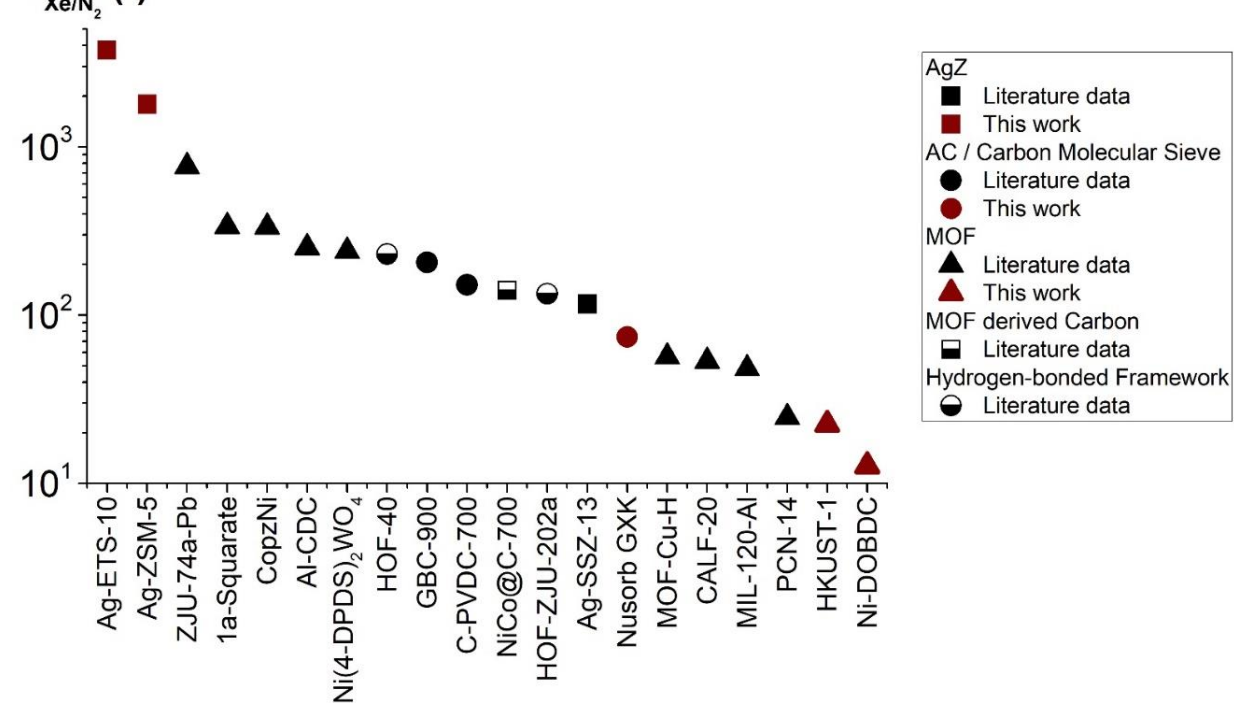
Xe collection from air at room temperature

- What about other materials in the literature ?
- Literature data: use of adsorption isotherm fits

Xe adsorption capacity (mol/kg) at 13.5 mPa



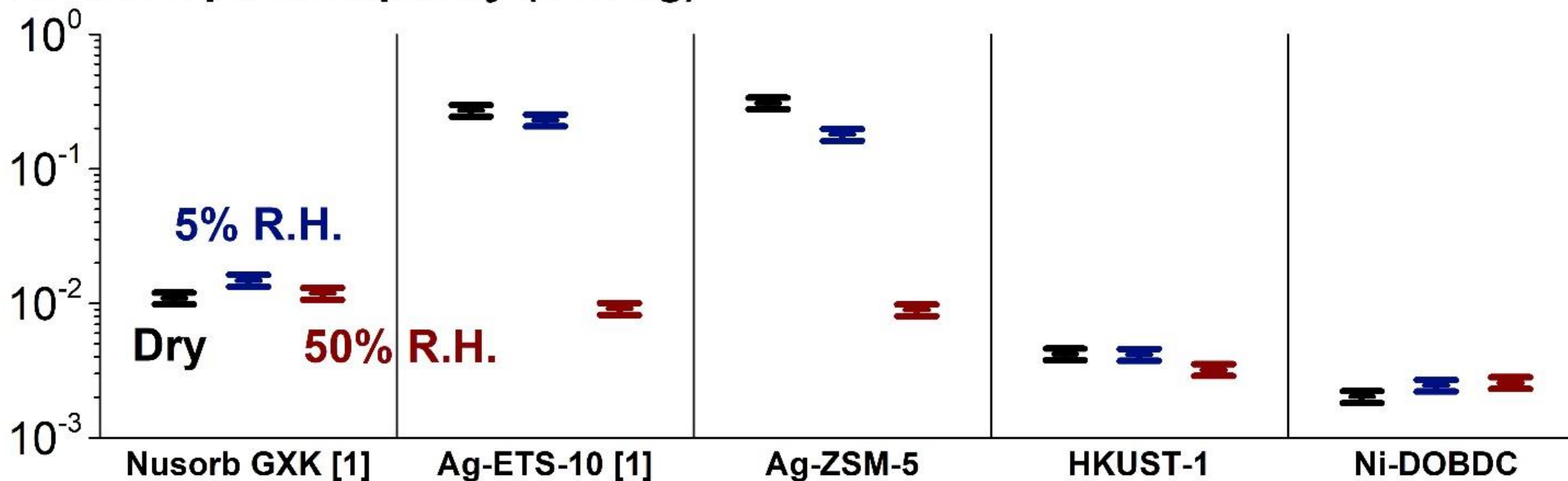
S_{Xe/N_2} (-)



Xe collection from air at room temperature

- What about moisture in the gas stream ?
 - Factor 30 decrease in Xe adsorption capacity on AgZs at 50% R.H.

Xe adsorption capacity (mol/kg)

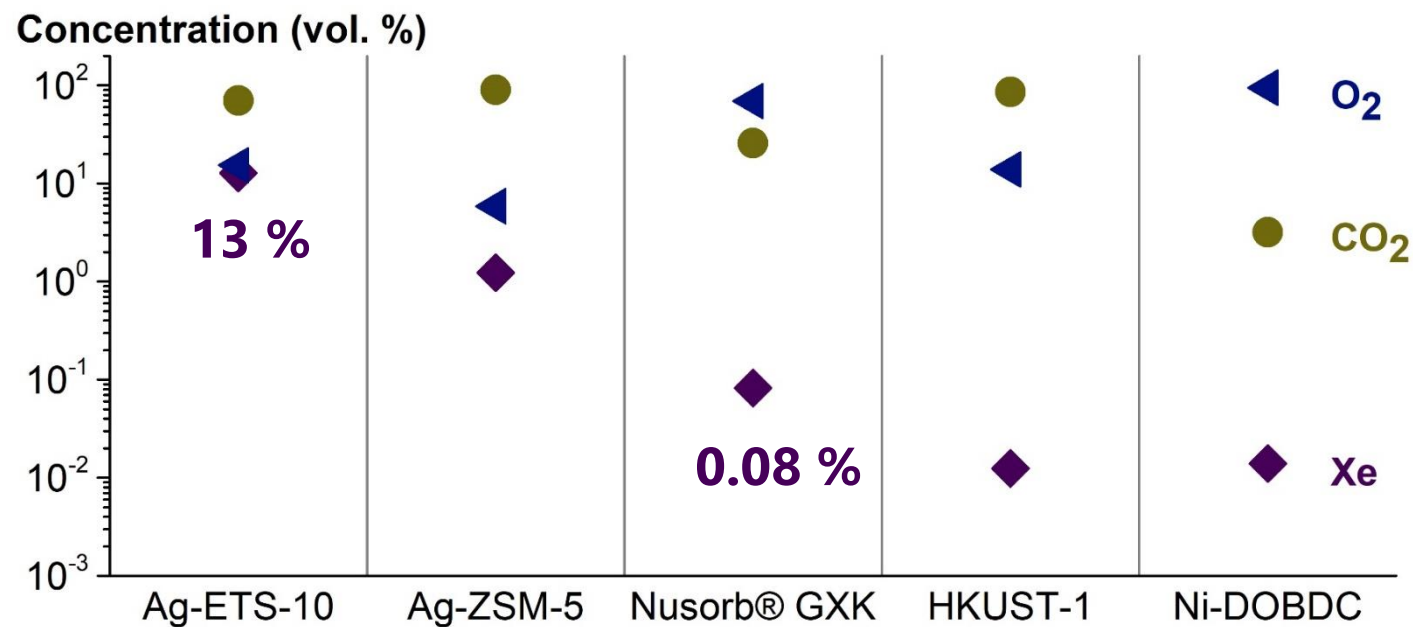


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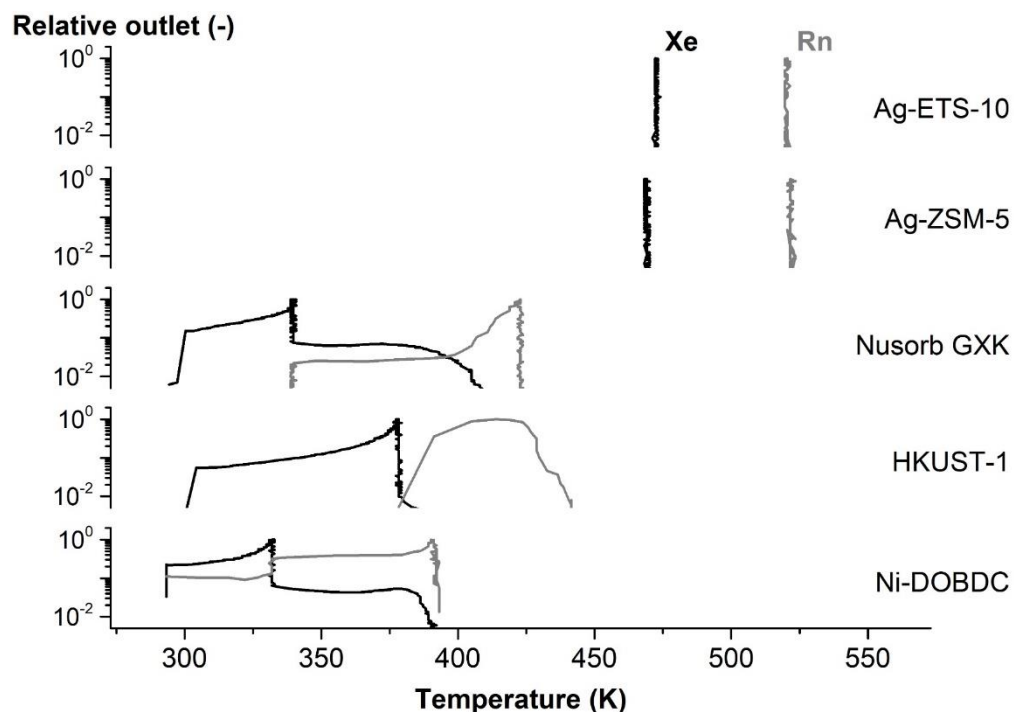
How easy and in which purity can we thermally recover the collected Xe from AgZs and MOFs compared to AC ?

Thermal Xe separation from air

- After single thermal cycle
 - 13 vol. % Xe recovered on Ag-ETS-10



- Impressive Xe/Rn separation on AgZs
 - **BUT** higher temperature required



Conclusions & perspectives

1. Xe adsorption & selectivity in air

- Unprecedented values on AgZs
- **NOT** in humid conditions

2. Thermal recovery of Xe from dry air

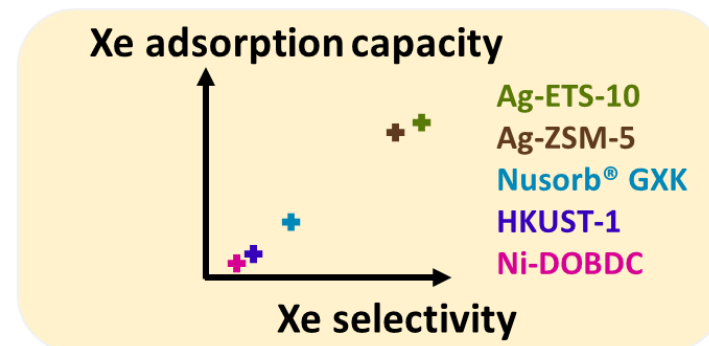
- Unprecedented values on AgZs
- **BUT** higher temperature than for MOFs/AC

- Future work: purity over multiple cycles, durability, other adsorbents ?

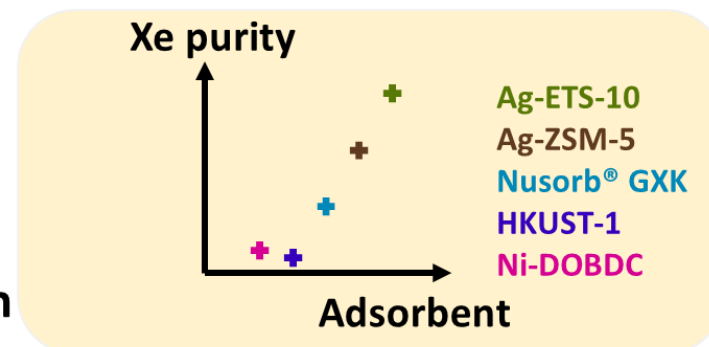
- Perspective: simple and low energy demanding systems

- Use for Xe measurements in emergency situations
- Use for Xe measurements for atmospheric transport studies

①
Xenon
collection



②
Xenon
purification





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Silver-exchanged zeolites for collecting and separating xenon directly from atmospheric air

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