



UHASSELT

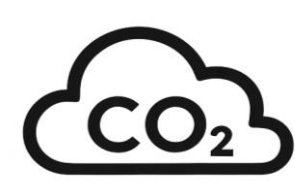
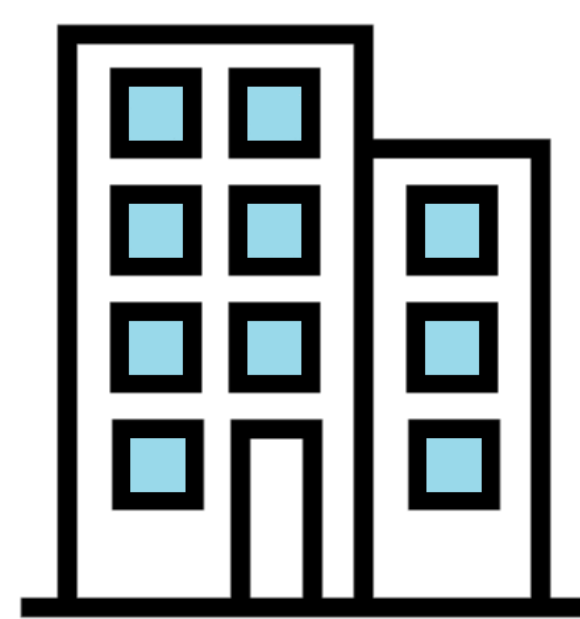
KNOWLEDGE IN ACTION

An In-Depth Study of the Thermodynamics and Kinetics of the Structural Phase Transition of Hydrothermally Synthesized W/VO₂ Microparticles

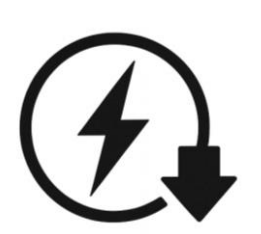
S. Cosemans^a, E. Heirman^a, L. Leufkens^{b,f}, M. A. Verheijen^{c,d}, K. Elen^{a,e}, D. Mann^{a,b,f}, A. Hardy^{a,e}, M. K. Van Bael^{a,e}, P. Buskens^{a,b,f}

^a Institute for Materials Research (imo-imomec), DESINE Group, Hasselt University, Agoralaan building D, 3590 Diepenbeek, (Belgium); ^b The Netherlands Organisation for Applied Scientific Research (TNO), High Tech Campus 25, 5656 AE Eindhoven, (The Netherlands); ^c Department of Applied Physics, Eindhoven University of Technology, 5600 MB Eindhoven, (The Netherlands); ^d Eurofins Materials Science, High Tech Campus 11, 5656 AE Eindhoven, (The Netherlands); ^e imec vzw, imec, Wetenschapspark 1, B-3590 Diepenbeek, (Belgium); ^f Brightlands Materials Center, Urmonderbaan 22, Geleen 6167 RD, The Netherlands.

Introduction

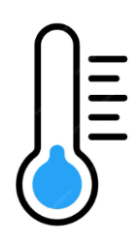


Buildings are responsible for about **1/3 of global energy use and CO₂ emissions**.



More than 50% of this energy used for heating and cooling. Where **windows alone account for over 30% of a building's total energy loss or gain.**

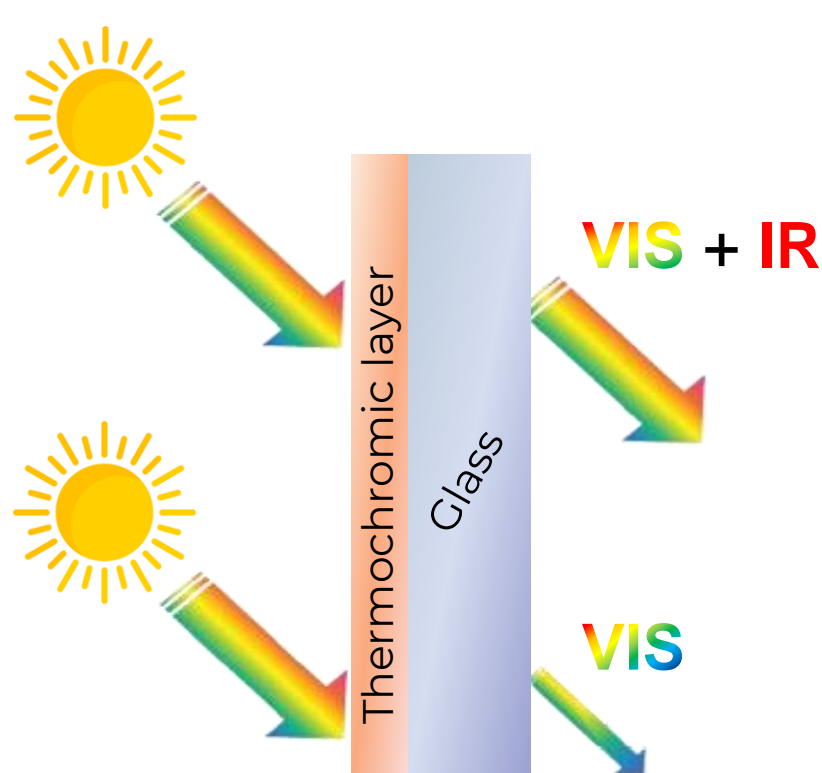
"Smart windows" with thermochromic VO₂ coatings offer a passive solution.



On a **cold day** the coating lets **solar infrared pass through**, providing passive heating.



On a **hot day** the coating **blocks solar infrared radiation**, passively cooling the buildings.



Switch temperature: ~20 °C (ambient)

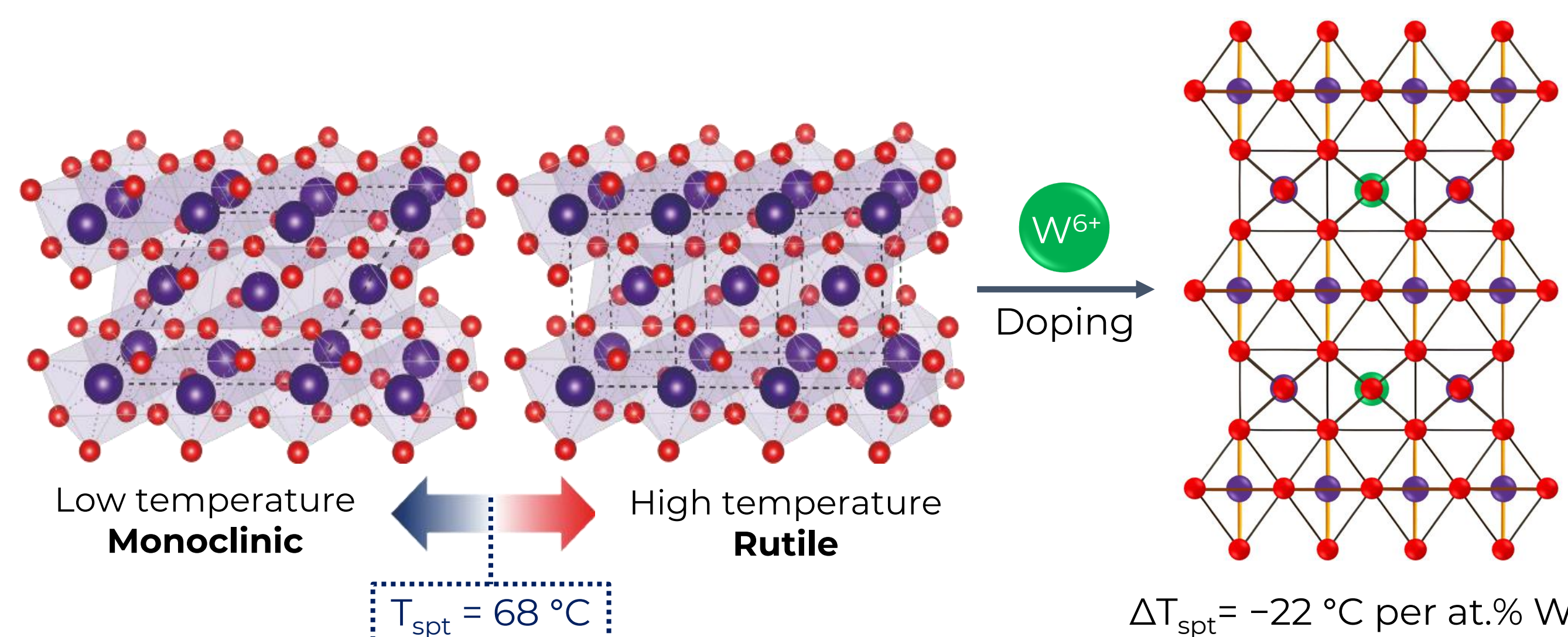
Research Aim

The natural phase transition temperature of VO₂ (~68 °C) is too high for smart window use. **Tungsten doping** lowers this temperature but also **alters the material's structure and transition dynamics**, which are still **not well understood for hydrothermally synthesized VO₂**.

This work aims to **reveal how W-doping and microstructure affect the thermodynamics and kinetics of the structural phase transition** in VO₂ microparticles made by a non-harmful, scalable hydrothermal method.

Theory

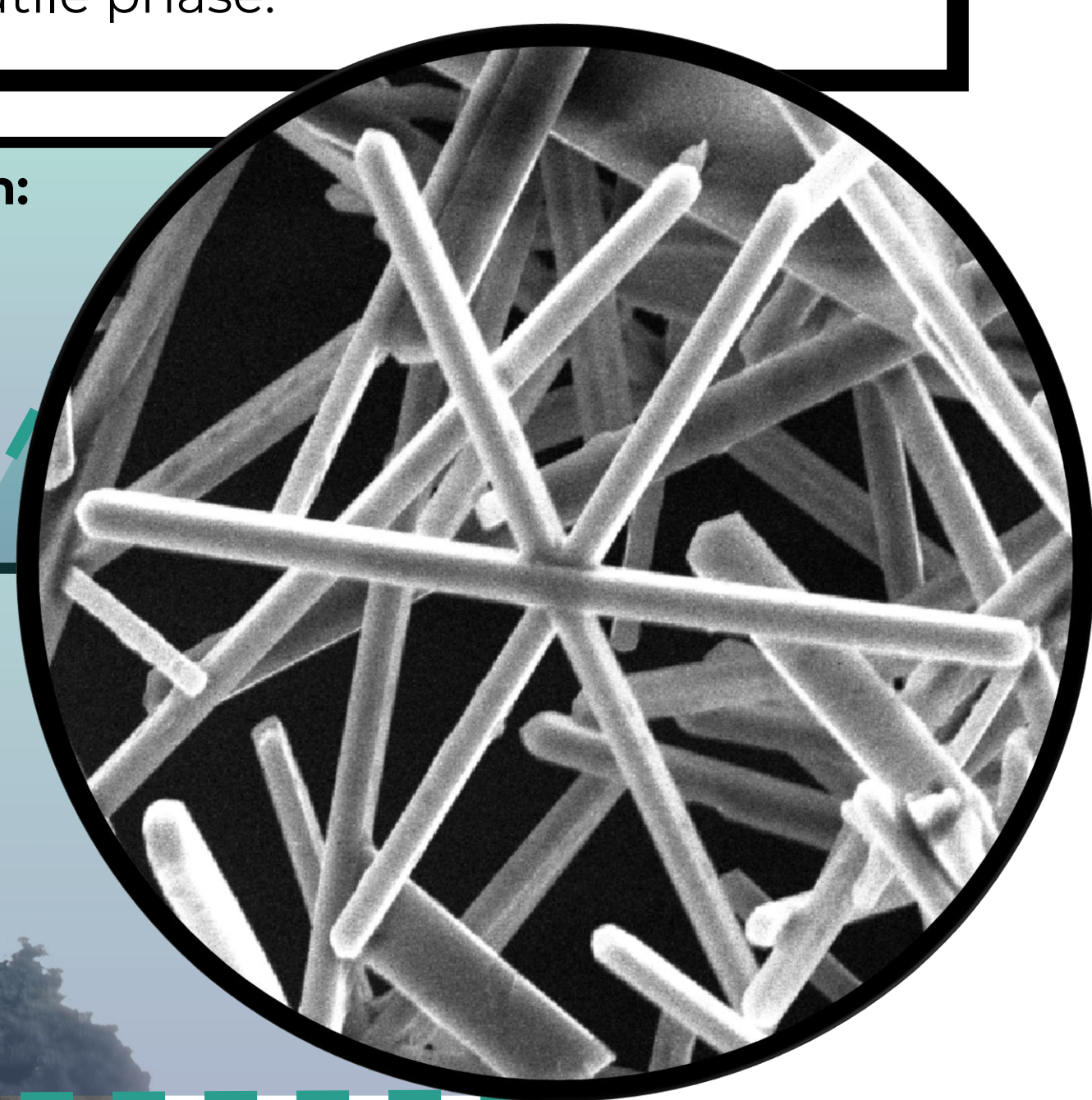
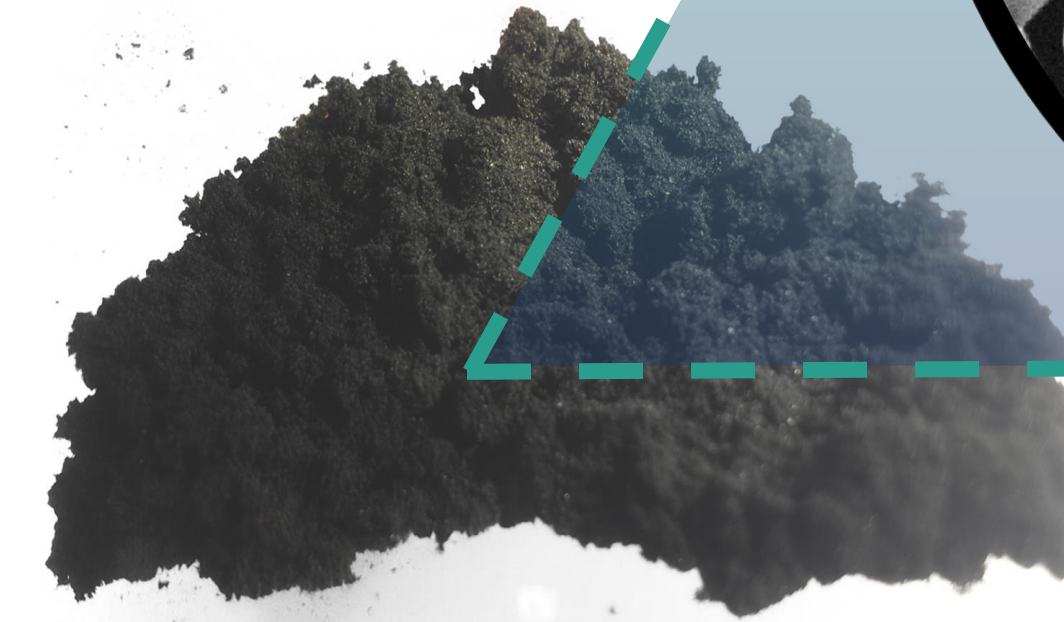
VO₂ structural phase transition (SPT)



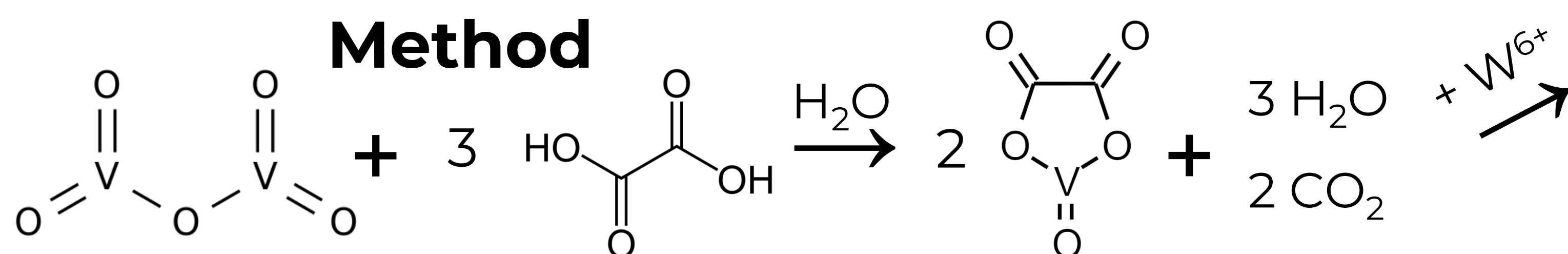
The SPT induces a change in the materials optical properties, from IR transparent to IR blocking. The temperature of the **SPT can be lowered using dopants such as tungsten**, as W⁶⁺ substitution for V⁴⁺ distorts the lattice and weakens V-V bonds, promoting the metallic rutile phase.

Hydrothermal reaction:

Reaction time: 72 h
Temperature: 230 °C
Dopant: Tungsten
Atom percentage W: 2.0 at%

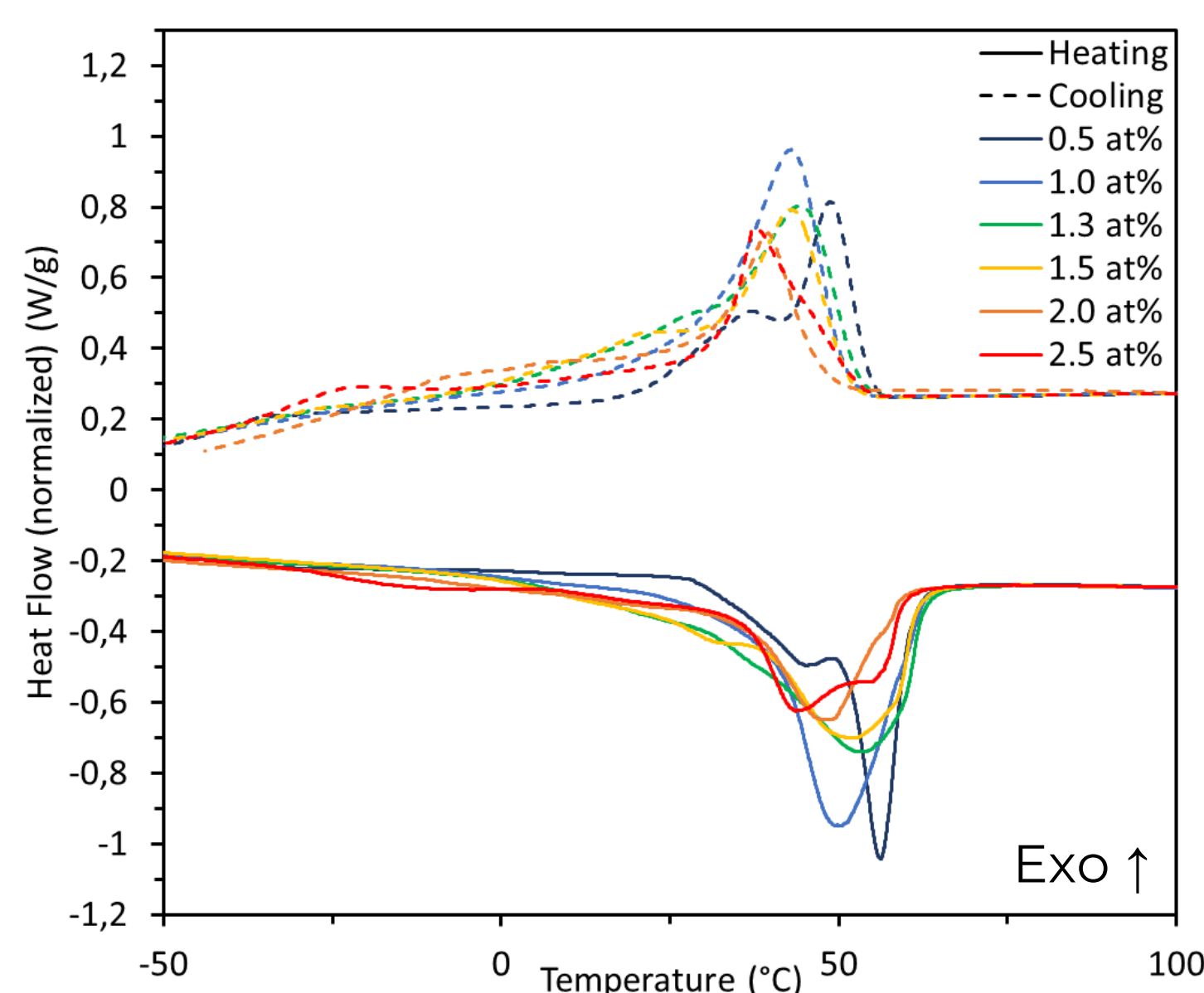


Method



Results and Discussion

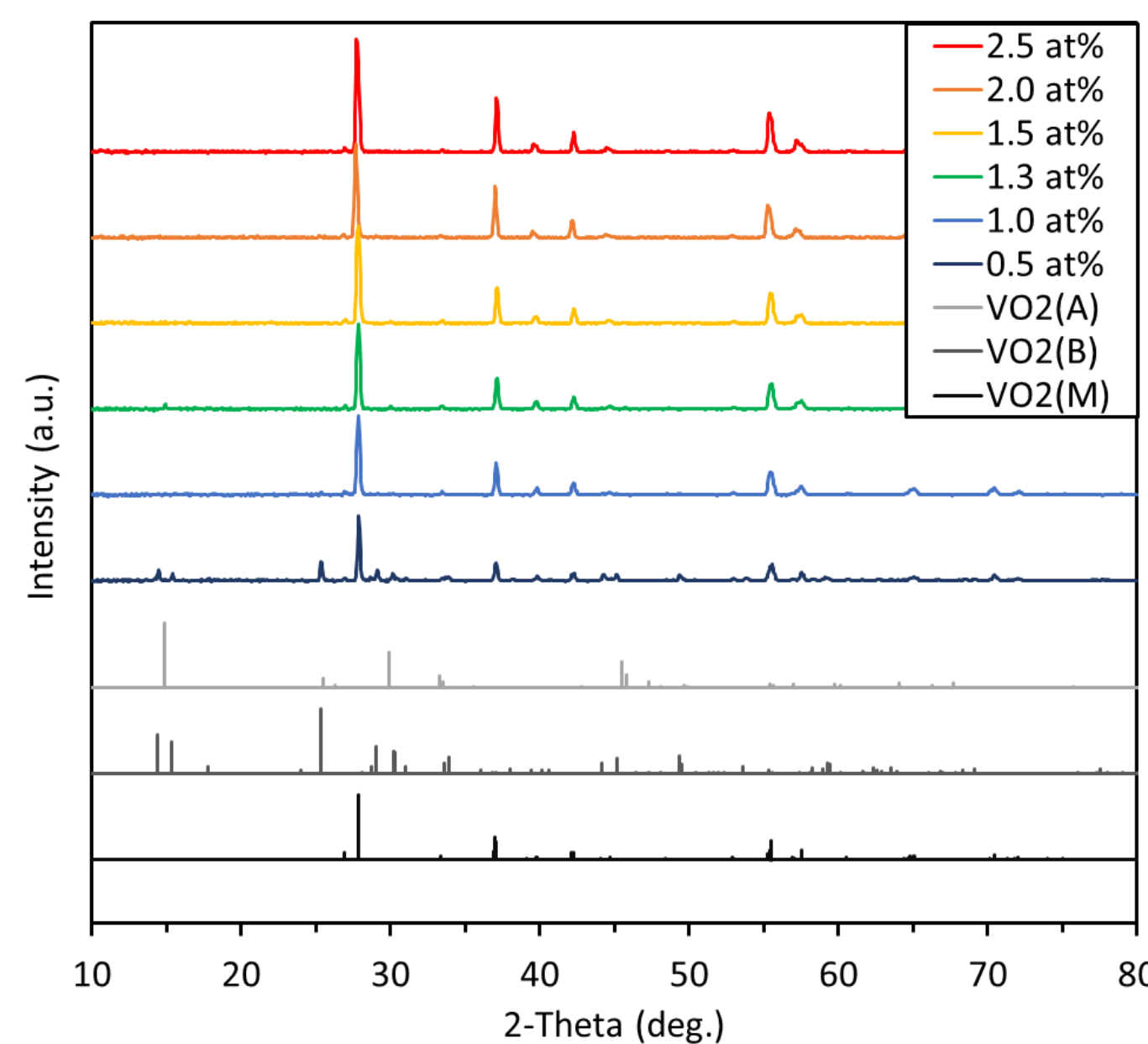
Importance of tungsten doping



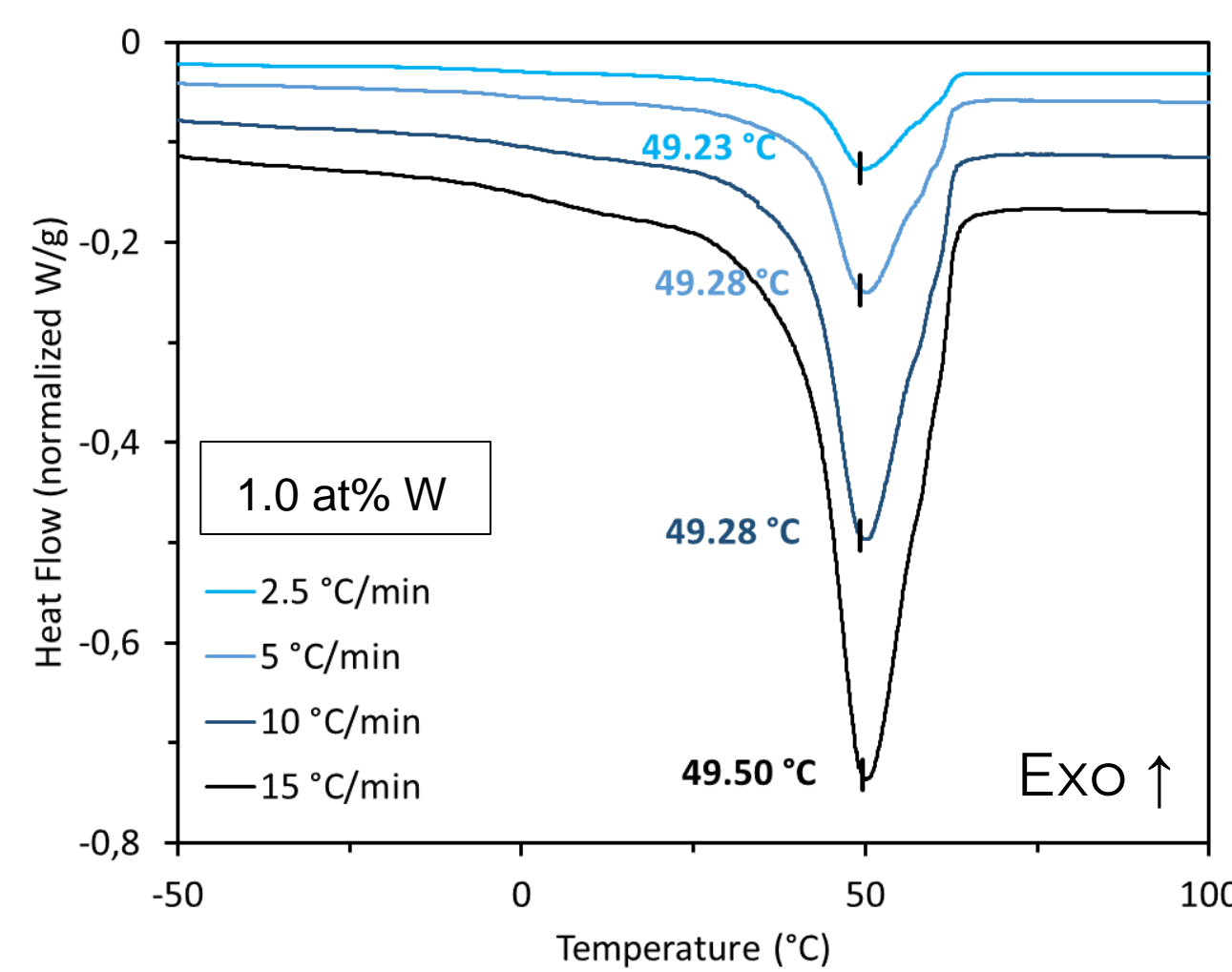
Increasing W content lowers the VO₂ switching temperature, though the reduction is smaller than expected due to:

- Inhomogeneous W incorporation:** 0.1-1.8 at% among individual crystals in the 1.5 at% W sample. (TEM-EDX)
- Non-uniform crystal size and shape:** rods (10.7 ± 2.3 μm), asterisks (arm length 12.8 ± 3.1 μm). (SEM)

W promotes crystallization of VO₂ in the monoclinic (M) phase. With more W, VO₂(B) peaks disappear and monoclinic peaks increase in intensity, confirming improved phase purity.



Kinetic analysis of the W-doped VO₂ phase transition



DSC was performed at multiple heating rates (2.5 – 15 °C min⁻¹) for W-doped VO₂.

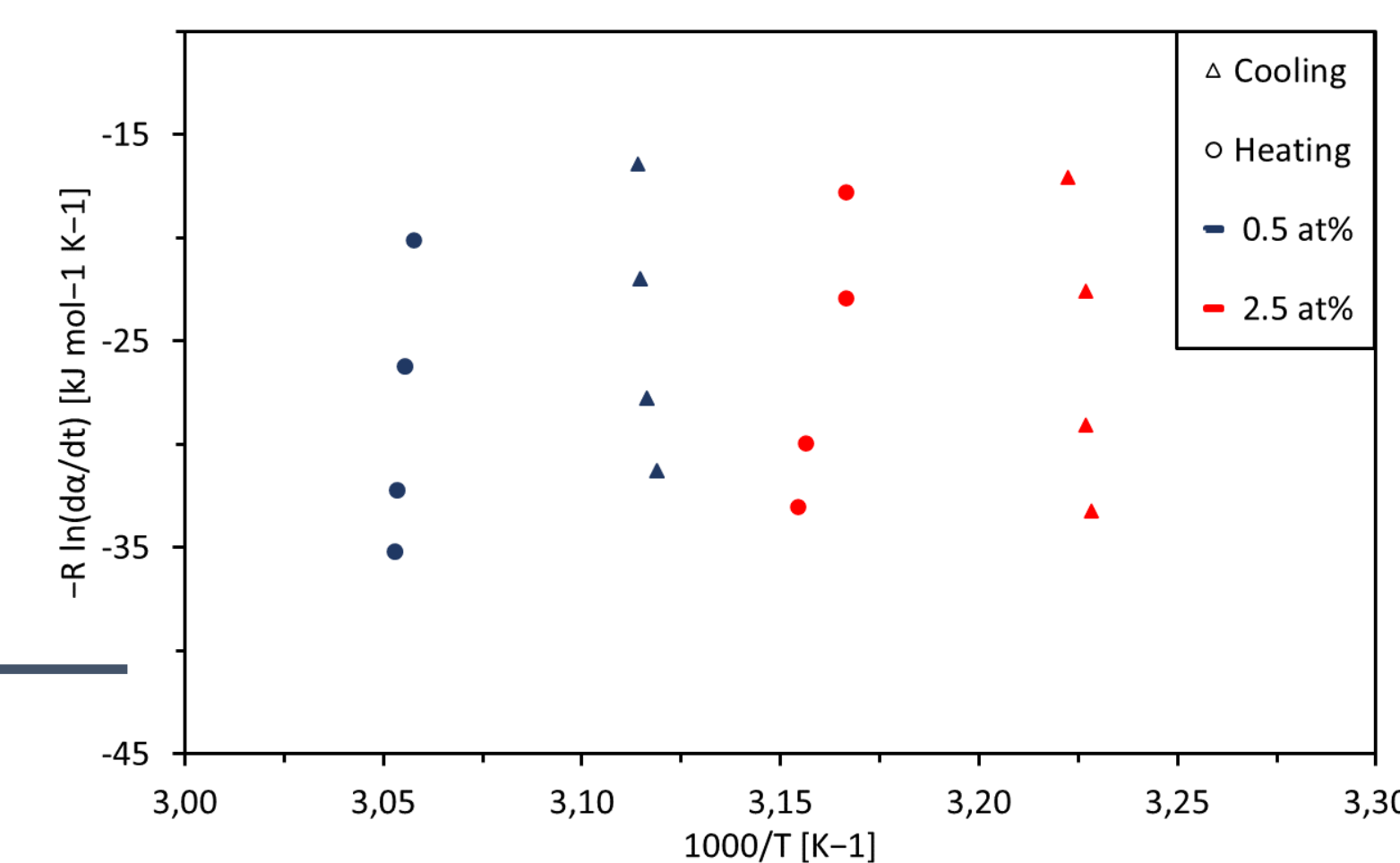
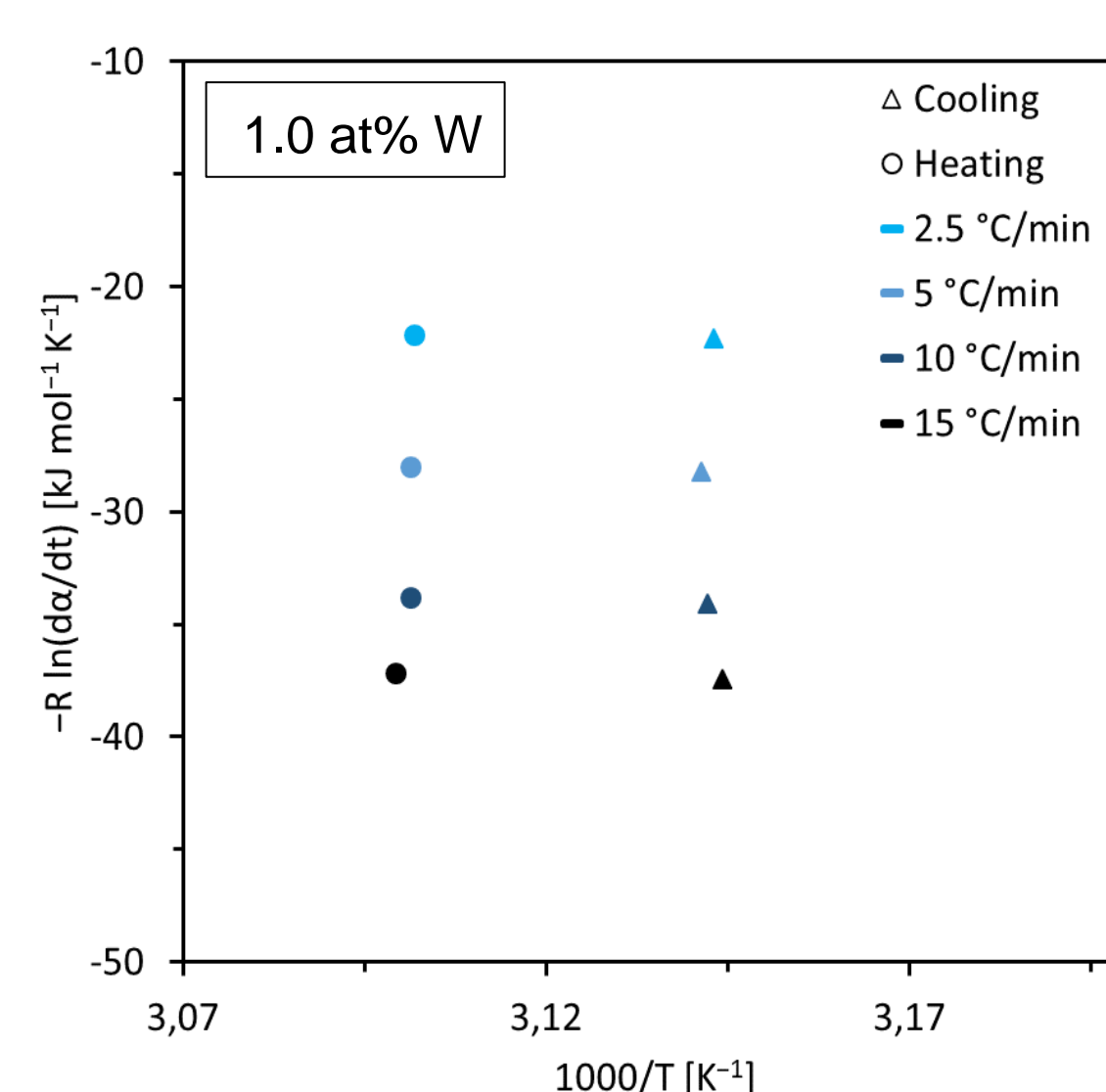
The running integral of the DSC signal was used to calculate conversion (α) vs T and time. At α = 0.5, the temperature (T_{50%}) and conversion rate (dα/dt) were extracted.

Using the Friedman isoconversional method, plots of ln(dα/dt) vs 1/T yielded the activation energy (E_a).

The E_a trend with temperature remains unchanged, but higher W content shifts T_{50%} to lower temperatures.

Friedman differential isoconversional method

$$-R \ln \left(\frac{d\alpha}{dt} \right)_{\alpha_i} = -R \ln(A_i f(\alpha_i)) + \frac{E_a}{T_{\alpha_i}}$$



Final Insights

- The developed hydrothermal synthesis yields **highly crystalline, phase-pure monoclinic VO₂**. W doping promotes the formation of VO₂(M) and lowers transition temperature.
- The **activation barrier/energy increases near the transition temperature**, causing phase switching to slow dramatically as the system approaches T_{spt}.
- Increasing W content shifts the 50% conversion point**, enabling control over the switching temperature to match different ambient climates or functional requirements.
- The asymptotic activation behavior shows that **nucleation**, rather than phase growth, is the **rate-determining step** for both of the phase transitions.

References

Mann D. et al., Energies, (2020), 13, 2842
Calvi L. et al., Sol. Energy Mater. Sol. Cells, (2022), 242, 111783
Timmers K. et al., Inorg. Chem., (2024), 63, 5400-5413

